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<b>(21) International Application Number:</b> PCT/US96/09471 <b>(22) International Filing Date:</b> 4 June 1996 (04.06.96) <b>(30) Priority Data:</b> 08/481,808 7 June 1995 (07.06.95) US <b>(71) Applicant:</b> PROTOTEK, INC. [US/US]; 6497 Sierra Lane, Dublin, CA 94568 (US). <b>(72) Inventors:</b> ZIMMERMAN, Mary, P.; 3727 Crofter's Court, Pleasanton, CA 94588 (US). SMITH, Robert, E.; 574 Escondido Circle, Livermore, CA 94550 (US). BECKER, Mark, W.; Apartment 76, 1390 Creekside Drive, Walnut Creek, CA 94596 (US). <b>(74) Agents:</b> THOMAS, Timothy, N. et al.; Woodard, Emhardt, Naughton, Moriarty & McNett, Bank One Center/Tower, Suite 3700, 111 Monument Circle, Indianapolis, IN 46204 (US).	<b>(81) Designated States:</b> AL, AM, AT, AU, AZ, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IL, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, UZ, VN, ARIPO patent (KE, LS, MW, SD, SZ, UG), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).  <b>Published</b> <i>With international search report.</i>	
<b>(54) Title:</b> $\alpha$ -(1,3-DICARBONYLENOL ETHER) METHYL KETONES AS CYSTEINE PROTEASE INHIBITORS <b>(57) Abstract</b> <p>Cysteine protease inhibitors which deactivate the protease by covalently bonding to the cysteine protease and releasing the enolate of a 1,3-dicarbonyl (or its enolic form). The cysteine protease inhibitors of the present invention accordingly comprise a first portion which targets a desired cysteine protease and positions the inhibitor near the thiolate anion portion of the active site of the protease, and a second portion which covalently bonds to the cysteine protease and irreversibly deactivates that protease by providing a carbonyl or carbonyl-equivalent which is attacked by the thiolate anion of the active site of the cysteine protease to sequentially cleave a <math>\beta</math>-dicarbonylenol ether leaving group.</p>		

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**$\alpha$ -(1,3-DICARBONYLENOL ETHER) METHYL KETONES  
AS CYSTEINE PROTEASE INHIBITORS**

Relation to Pending Applications

This application is a continuation-in-part of U.S. Patent  
5 application Serial No. 08/164,031, filed December 8, 1993.

FIELD OF THE INVENTION

The present invention relates generally to cysteine  
protease inhibitors, and more particularly to cysteine  
protease inhibitors which are peptidyl ketones which contain  
10 dicarbonyl enolether leaving groups. The cysteine protease  
inhibitors of the present invention are particularly designed  
for the in vivo management of cysteine proteases,  
particularly cathepsins B, L, H and C, calpains I and II,  
interleukin 1- $\beta$ -converting enzyme ("ICE"), and the primitive  
15 enzymatic counterparts of these cysteine proteases.

BACKGROUND TO THE INVENTION

Cysteine proteases associated with human disease states  
can be grouped into three categories: (1) lysosomal  
cathepsins; (2) cytosolic calpains and processing enzymes  
20 such as interleukin converting enzymes; and (3) prokaryotic  
enzymes with autocatalytic activation. Cathepsins B, H, and  
L are cysteinyl proteases involved in normal protein  
degradation. As such, they are generally located in the  
lysosomes of cells. When these enzymes are found  
25 extralysosomally they have been implicated by use of synthetic  
substrate technology and by natural endogenous inhibitors as  
playing a causative role in a number of disease states such

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as rheumatoid arthritis, osteo arthritis, pneumocystis carinii, schistosomiasis, trypanosoma cruzi, trypanosoma brucei brucei, Crithidia fusiculata, malaria, periodontal disease, tumor metastasis, metachromatic leukodystrophy, muscular dystrophy, etc. For example, a connection between cathepsin B-type enzymes and rheumatoid arthritis has been suggested in van Noorden and Everts, "Selective Inhibition of Cysteine Proteinases by Z-Phe-Ala-CH<sub>2</sub>F Suppresses Digestion of Collagen by Fibroblasts and Osteoclasts," 178 Biochemical and Biophysical Research Communications 178; Rifkin, Vernillo, Kleekner, Auszmann, Rosenberg and Zimmerman, "Cathepsin B and L Activities in Isolated Osteoclasts," 179 Biochemical and Biophysical Research Communications 63; Grinde, "The Thiol Proteinase Inhibitors, Z-Phe-Phe-CHN<sub>2</sub> and Z-Phe-Ala-CHN<sub>2</sub>, Inhibit Lysosomal Protein Degradation in Isolated Rat Hepatocytes," 757 Biochimica et Biophysica Acta 15; Mason, Bartholomew and Hardwick, "The Use of Benzyloxycarbonyl [<sup>125</sup>I]iodotyrosylalanyldiazomethane as a Probe for Active Cysteine Proteinases in Human Tissues," 263 Biochem. J. 945; van Noorden, Smith and Rasnick, "Cysteine Proteinase Activity in Arthritic Rat Knee Joints and the Effects of a Selective Systemic Inhibitor, Z-Phe-Ala-CH<sub>2</sub>F," 15 J. Rheumatol. 1525; and van Noorden, Vogels and Smith, "Localization and Cytophotometric Analysis of Cathepsin B Activity in Unfixed and Undecalcified Cryostat Sections of Whole Rat Knee Joints," 37 J. Histochemistry and Cytochemistry 617. A connection between cathepsin B and osteo arthritis has been suggested in Delaissé, Eeckhout and Vaes, "In Vivo and In Vitro Evidence for the Involvement of Cysteine Proteinases in Bone Resorption," 125 Biochemical and Biophysical Research Communications 441; a connection between cathepsin B and pneumocystis carinii has been suggested in Hayes, Stubberfield, McBride and Wilson, "Alterations in Cysteine Proteinase Content of Rat Lung Associated with Development of Pneumocystis Carinii Infection," 59 Infection

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and Immunity 3581; a connection between cysteine proteinases and schistosomiasis has been suggested in Cohen, Gregoret, Amiri, Aldape, Railey and McKerrow, "Arresting Tissue Invasion of a Parasite by Protease Inhibitors Chosen With the Aid of Computer Modeling," 30 Biochemistry 11221. A connection between cysteine proteinases and trypanosoma cruzi, trypanosoma brucei brucei and crithidia fasciculata has been suggested in Ashall, Harris, Roberts, Healy and Shaw, "Substrate Specificity and Inhibitor Sensitivity of a Trypanosomatid Alkaline Peptidase," 1035 Biochimica et Biophysica Acta 293, and/or in Ashall, Angliker and Shaw, "Lysis of Trypanosomes by Peptidyl Fluoromethyl Ketones," 170 Biochemical and Biophysical Research Communications 923. A connection between cysteine proteinases and malaria has been suggested in Rosenthal, Wollish, Palmer and Rasnick, "Antimalarial Effects of Peptide Inhibitors of a Plasmodium Falciparum Cysteine Proteinase," 88 J. Clin. Invest. 1467, and in Rosenthal, Lee and Smith, "Inhibition of a Plasmodium Vinckei Cysteine Proteinase Cures Murine Malaria," 91 J. Clin. Invest. 1052. A connection between cathepsin B and tumor metathesis has been suggested in Smith, Rasnick, Burdick, Cho, Rose and Vahratian, "Visualization of Time-Dependent Inactivation of Human Tumor Cathepsin B Isozymes by a Peptidyl Fluoromethyl Ketone Using a Fluorescent Print Technique," 8 Anti-cancer Research 525. A connection between cathepsin B and cancer has been suggested in Gordon and Mourad, 2 Blood Coagulation and Fibrinolysis 735. A connection between cathepsin B and periodontal disease has been suggested in Cox, Cho, Eley and Smith, "A Simple, Combined Fluorogenic and Chromogenic Method for the Assay of Proteases in Gingival Crevicular Fluid," 25 J. Periodont. Res. 164; Uitto, Larjava, Heino and Sorsa, "A Protease of Bacteroides Gingivalis Degrades Cell Surface and Matrix Glycoproteins of Cultured Gingival Fibroblasts and Induces Secretion of Collagenase and Plasminogen Activator,"

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- 57 Infection and Immunity 213; Kunimatsu, Yamamoto, Ichimaru, Kato and Kato, "Cathepsins B, H and L Activities in Gingival Crevicular Fluid From Chronic Adult Periodontitis Patients and Experimental Gingivitis Subjects," 25 J Periodont Res 69;
- 5 Beighton, Radford and Naylor, "Protease Activity in Gingival Crevicular Fluid From Discrete Periodontal Sites in Humans With Periodontitis or Gingivitis"; 35 Archs oral Biol. 329; Cox and Eley, "Preliminary Studies on Cysteine and Serine Proteinase Activities in Inflamed Human Gingiva Using
- 10 Different 7-Amino-4-Trifluoromethyl Coumarin Substrates and Protease Inhibitors," 32 Archs oral Biol. 599; and Eisenhauer, Hutchinson, Javed and McDonald, "Identification of a Cathepsin B-Like Protease in the Crevicular Fluid of Gingivitis Patients," 62 J Dent Res 917. A connection
- 15 between cathepsin B and metachromatic leukodystrophy has been suggested in von Figura, Steckel, Conary, Hasilik and Shaw, "Heterogeneity in Late-Onset Metachromatic Leukodystrophy. Effect of Inhibitors of Cysteine Proteinases," 39 Am J Hum Genet. 371; a connection between cathepsin B and muscular
- 20 leukodystrophy has been suggested in Valentine, Winand, Pradhan, Moise, de Lahunta, Kornegay and Cooper, "Canine X-Linked Muscular Dystrophy as an Animal Model of Duchenne Muscular Dystrophy: A Review," 42 Am J Hum Genet 352; a connection between cathepsin B and rhinovirus has been
- 25 suggested in Knott, Orr, Montgomery, Sullivan and Weston, "The Expression and Purification of Human Rhinovirus Protease 3C," 182 Eur. J. Biochem. 547; a connection between cathepsin B and kidney disease has been suggested in Baricos, O'Connor, Cortez, Wu and Shah, "The Cysteine Proteinase Inhibitor,
- 30 E-64, Reduces Proteinuria in an Experimental Model of Glomerulonephritis," 155 Biochemical and Biophysical Research Communications 1318; and a connection between cathepsin B and multiple sclerosis has been suggested in Dahlman, Rutschmann, Kuehn and Reinauer, "Activation of the Multicatalytic
- 35 Proteinase from Rat Skeletal Muscle by Fatty Acids or Sodium Dodecyl Sulphate," 228 Biochem. J. 171.

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Connections between certain disease states and cathepsins H and C have also been established. For example, cathepsin H has been directly linked to the causative agents of Pneumocystis carinii and in the neuromuscular diseases Duchenne dystrophy, polymyositis, and neurogenic disorders. Stauber, Riggs and Schochet, "Fluorescent Protease Histochemistry in Neuromuscular Disease," Neurology 194 (Suppl. 1) March 1984; Stauber, Schochet, Riggs, Gutmann and Crosby, "Nemaline Rod Myopathy: Evidence for a Protease Deficiency," Neurology 34 (Suppl. 1) March 1984. Similarly, cathepsin C has been directly linked to muscular diseases such as nemaline myopathy, to viral infections, and to processing and activation of bone marrow serine proteases (elastase and granzyme A). McGuire, Lipsky and Thiele, "Generation of Active Myeloid and Lymphoid Granule Serine Proteases Requires Processing by the Granule Thiol Protease Dipeptidyl Peptidase I, 268 J. Biol. Chem. 2458-67; L. Polgar, Mechanisms of Protease Action (1989); Brown, McGuire and Thiele, "Dipeptidyl Peptidase I is Enriched in Granules of In Vitro- and In Vivo-Activated Cytotoxic T Lymphocytes," 150 Immunology 4733-42. The Brown et al. study effectively demonstrated the feasibility of inhibiting cathepsin C (DPP-I) in the presence of other cysteinyl enzymes based on substrate specificity. Unfortunately, the diazoketones used in that study are believed to be mutagenic and not appropriate for in vivo application.

The cytosolic or membrane-bound cysteine proteases called calpains have also been implicated in a number of disease states. For example, calpain inhibitor can be useful for the treatment of muscular disease such as muscular dystrophy, amyotrophy or the like, 25 Taisha (Metabolism) 183 (1988); 10 J. Pharm. Dynamics 678 (1987); for the treatment of ischemic diseases such as cardiac infarction, stroke and the like, 312 New Eng. J. Med. 159 (1985); 43 Salshin Igaku 783 (1988); 36 Arzneimittel Forschung/Drug Research 190, 671

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(1986); 526 Brain Research 177 (1990); for improving the consciousness disturbance or motor disturbance caused by brain trauma, 16 Neurochemical Research 483 (1991); 65 J. Neurosurgery 92 (1986); for the treatment of diseases caused by the demyelination of neurocytes such as multiple sclerosis, peripheral nervous neuropathy and the like, 47 J. Neurochemistry 1007 (1986); and for the treatment of cataracts, 28 Investigative Ophthalmology & Visual Science 1702 (1987); 34 Experimental Eye Research 413 (1982); 6 Lens and Eye Toxicity Research 725 (1989); 32 Investigative Ophthalmology & Visual Science 533 (1991).

Calpain inhibitors may also be used as therapeutic agents for fulminant hepatitis, as inhibitors against aggregation of platelet caused by thrombin, 57 Thrombosis Research 847 (1990); and as a therapeutic agent for diseases such as breast carcinoma, prostatic carcinoma or prostatomegaly, which are suspected of being caused by an abnormal activation of the sex hormone receptors.

Certain protease inhibitors have also been associated with Alzheimer's disease. See, e.g., 11 Scientific American 40 (1991). Further, thiol protease inhibitors are believed to be useful as anti-inflammatory drugs, 263 J. Biological Chem. 1915 (1988); 98 J. Biochem. 87 (1985); as antiallergic drugs, 42 J. Antibiotics 1362 (1989); and to prevent the metastasis of cancer, 57 Seikagaku 1202 (1985); Tumor Progression and Markers 47 (1982); and 256 J. Biological Chemistry 8536 (1984).

Further, Interleukin 1- $\beta$ -Converting Enzyme (ICE) has been shown to be a cysteine protease implicated in the formation of the cytokine IL-1 $\beta$  which is a potent mediator in the pathogenesis of chronic and acute inflammatory diseases. Tocci and Schmidt, ICOP Newsletter, September 1994. Inhibitors to this enzyme have recently been reported, including Thornberry, Peterson, Zhao, Howard, Griffin, and Chapman, "Inactivation of Interleukin-1 $\beta$ -Converting Enzyme by



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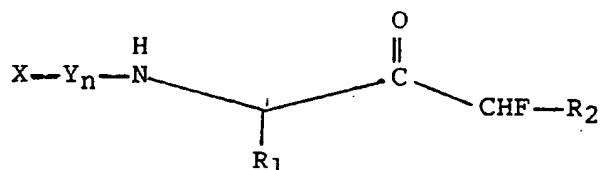
Peptide (Acyloxy)methyl Ketones, 33 Biochemistry 3934 (1994);  
Dolle, Singh, Rinker, Hoyer, Prasad, Graybill, Salvino,  
Helaszek, Miller and Ator, "Aspartyl  
 $\alpha$ -((1-Phenyl-3-(trifluoromethyl)-pyrazol-5-yl)-oxy)methyl  
5 Ketones as Interleukin-1 $\beta$  Converting Enzyme Inhibitors:  
Significance of the P<sub>1</sub> and P<sub>3</sub> Amido Nitrogens for  
Enzyme-Peptide Inhibitor Binding" 37 J. Med. Chem. 3863;  
Mjalli, Chapman, MacCoss, Thornberry, Peterson, "Activated  
Ketones as Potent Reversible Inhibitors of Interleukin-1 $\beta$ -  
10 Converting Enzyme" 4 Bioorganic & Medicinal Chemistry Letters,  
1965; and Dolle, Singh, Whipple, Osifo, Speier, Graybill,  
Gregory, Harris, Helaszek, Miller and Ator "Aspartyl  
 $\alpha$ -((Diphenylphosphinyl)-oxy)-methyl Ketones as Novel  
Inhibitors of Interleukin-1 $\beta$ -Converting Enzyme: Utility of  
15 the Diphenylphosphonic Acid Leaving Group for the Inhibition  
of Cysteine Proteases" 38 J. Med. Chem. 220.

The most promising type of cysteine proteinase inhibitors  
have an activated carbonyl with a suitable  $\alpha$ -leaving group  
fused to a programmed peptide sequence that specifically  
20 directs the inhibitor to the active site of the targeted  
enzyme. Once inside the active site, the inhibitor carbonyl  
is attacked by a cysteine thiolate anion to give the  
resulting hemiacetal, which collapses via a 1,2-thermal  
migration of the thiolate and subsequent displacement of the  
25  $\alpha$ -keto-leaving group. The bond between enzyme and  
inhibitor is then permanent and the enzyme is irreversibly  
inactivated.

The usefulness of an inhibitor in inactivating a  
particular enzyme therefore depends not only on the "lock and  
30 key" fit of the peptide portion, but also on the reactivity  
of the bond holding the  $\alpha$ -leaving group to the rest of the  
inhibitor. It is important that the leaving group be  
reactive only to the intramolecular displacement via a  
1,2-migration of sulfur in the breakdown of the  
35 hemithioacetal intermediate.

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Groundbreaking work regarding cysteine proteinase inhibitors having an activated carbonyl, a suitable  $\alpha$ -leaving group and a peptide sequence that effectively and specifically directs the inhibitor to the active site of the targeted enzyme was disclosed in U.S. Patent 4,518,528 to Rasnick, incorporated herein by reference. That patent established peptidyl fluoromethyl ketones to be unprecedented inhibitors of cysteine proteinase in selectivity and effectiveness. The fluoromethyl ketones described and synthesized by Rasnick included those of the formula:



wherein  $\text{R}_1$  and  $\text{R}_2$  are independently selected from the group hydrogen, alkyl of 1-6 carbons, substituted alkyl of 1-6 carbons, aryl, and alkylaryl where the alkyl group is of 1-4 carbons;  $n$  is an integer from 1-4 inclusive;  $\text{X}$  is a peptide end-blocking group; and  $\text{Y}$  is an amino acid or peptide chain of from 1-6 amino acids.

Peptidylketone inhibitors using a phenol leaving group are similar to the peptidyl fluoroketones. As is known in the art, oxygen most closely approaches fluorine in size and electronegativity. Further, when oxygen is bonded to an aromatic ring these values of electronegativity become even closer due to the electron withdrawing effect of the  $\text{sp}^2$  carbons. The inductive effect of an  $\alpha$ -ketophenol versus an  $\alpha$ -ketofluoride when measured by the  $\text{pK}_a$  of the  $\alpha$ -hydrogen, appears comparable within experimental error.

Unfortunately, the leaving groups of prior art inhibitors that use a phenoxy group present problems of toxicity, solubility, etc. Solubility is of particular importance in

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the field of peptide derived drugs where bioavailability becomes the major criterion for the success of a drug. The solubility recommendation of the FDA is 5 mg/mL. Successful in vivo utility of prior art inhibitors has been limited due to the insolubility of the leaving groups. In vivo application to date has centered on inhibitors with peptide requirements allowing ester, acid or free amine side chains as those required in the inhibition of Interleukin-1 $\beta$ -converting enzyme: Revesz, Briswalter, Heng, Leutwiler, Mueller and Wuethrich, "35 Tetrahedron Letters 9693.

International application WO 93/09135 disclosed inhibitors again designed for Interleukin-1 $\beta$ -converting enzyme where an N-hydroxytetrazole was disclosed as a leaving group. Further, tetrazoles have also been used in other pharmaceutical products such as Ceforanide, etc.

The in vivo inhibition of other cysteine proteases using oxygen anionic leaving groups was first disclosed by Zimmerman, Bissell, and Smith in U.S. Patent 5,374,623 where it was disclosed that bioavailability is enhanced by the use of peptidyl  $\alpha$ -aromatic ether methyl ketones with selective peptide combinations not requiring the presence of a free amine or acid side chain. Later, a peptidyl (acyloxy)methyl ketone with lysine in the side chain was reported to have in vivo efficacy: Wagner, Smith, Coles, Copp, Ernest and Krantz, "In Vivo Inhibition of Cathepsin B by Peptidyl (Acyloxy)methyl Ketones," 37 J. Med. Chem. 1833. Unfortunately, peptidyl (acyloxy)methyl ketones are esters that are also subject to cleavage by esterases which makes the  $\alpha$ -ketoethers the preferred construction for cysteine protease inhibitors.

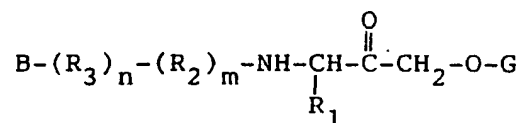
It can be seen from the foregoing that a need continues to exist for cysteine protease inhibitors with improved solubility and toxicity profiles, and which are particularly suitable for in vivo use. The present invention addresses that need.

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SUMMARY OF THE INVENTION

Briefly describing the present invention, there is provided a class of cysteine protease inhibitors which deactivates the protease by covalently bonding to the cysteine protease and releasing the enolate of a 1,3-dicarbonyl (or its enolic form). The cysteine protease inhibitors of the present invention accordingly comprise a first portion which targets a desired cysteine protease and positions the inhibitor near the thiolate anion portion of the active site of the protease, and a second portion which covalently bonds to the cysteine protease and irreversibly deactivates that protease by providing a carbonyl or carbonyl-equivalent which is attacked by the thiolate anion of the active site of the cysteine protease to sequentially cleave a  $\beta$ -dicarbonyl enol ether leaving group.

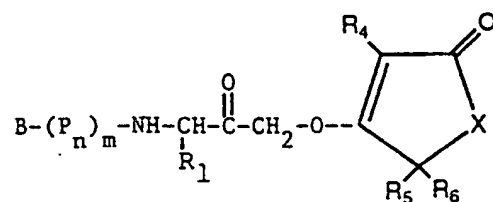
The cysteine protease inhibitors of the present invention may be defined by the formula below:



where

- B is H or an N-terminal blocking group;
- $R_{1-3}$  are the amino acid side chains of the  $P_{1-3}$  amino acids, respectively;
- n is 0 or 1;
- m is 0 or 1; and
- G is a five- or six-membered ring portion of the  $\beta$ -dicarbonyl enol ether leaving group as defined by the formulas below.

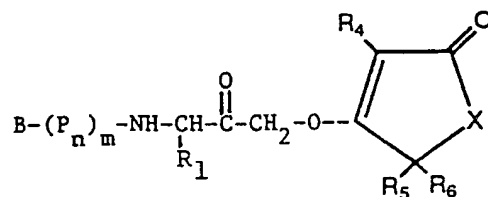
In one embodiment, the compositions of the present invention are cathepsin or calpain inhibitors of the formula:



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- where B is H or an N-terminal blocking group;  
 $R_1$  is the amino acid side chain of the  $P_1$  amino acid residue; wherein the  $P_1$  amino acid is not Asp;  
 each  $P_n$  is an amino acid residue, or is a  
 5 heterocyclic replacement of the amino acid wherein the heterocycle is a piperazine, a decahydroisoquinoline, a pyrrolinone, a pyridine, a pyridone, a carbolinone, a quinazoline, a pyrimidone or the like;  
 10 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxy, acyl, hydrogen, alkyl or phenyl;  
 $R_5$  and  $R_6$  are jointly a carboxyl group or a double bond terminating in an alkyl or an aryl  
 15 group, or are independently acyl, aryl or heteroaryl if  $R_4$  is hydrogen, alkyl or phenyl, or are independently acyl, alkyl, hydrogen, aryl or heteroaryl otherwise; and  
 X is N, S, O or  $CH_2$ .

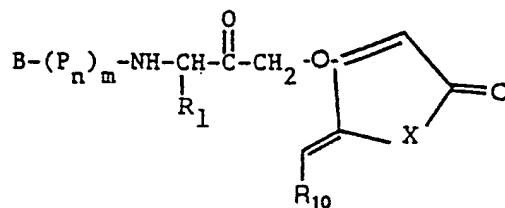
- 20 In another embodiment, the compositions of the present invention are ICE inhibitors of the formula:



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where B is H or an N-terminal blocking group;  
 $R_1$  is the Asp amino acid side chain;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 5 the heterocycle is a piperazine, a  
 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 10  $R_4$  is a hydroxyl, alkoxyl, acyl, hydrogen, alkyl  
 or phenyl;  
 $R_5$  and  $R_6$  are jointly a carboxyl group or a  
 double bond terminating in an alkyl or an aryl  
 group, or are independently acyl, aryl or heteroaryl  
 15 if  $R_4$  is hydrogen, alkyl or phenyl, or are  
 independently acyl, alkyl, hydrogen, aryl or  
 heteroaryl otherwise; and  
 X is N, S, O or  $CH_2$ .

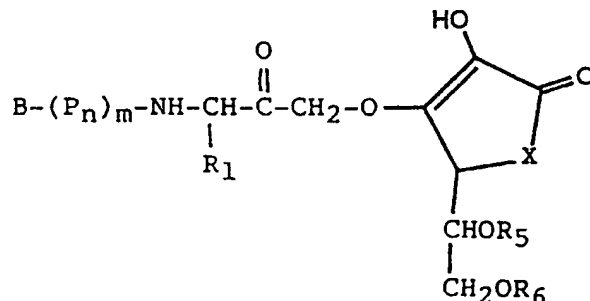
In another embodiment, the cysteine protease inhibitor is  
 20 of the formula:



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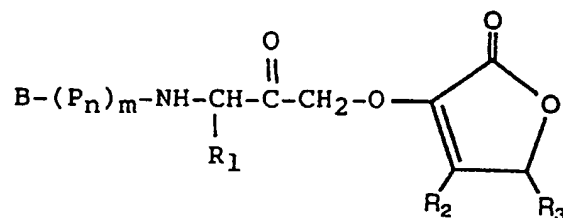
where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 5 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxy, acyl, hydrogen, alkyl  
 10 or phenyl;  
 $R_{10}$  is H or an optionally substituted alkyl, aryl,  
 heteroaryl, or the residue of a sugar; and  
 X is N, S, O or  $CH_2$ .

In another embodiment, the cysteine protease inhibitor is  
 15 of the formula:



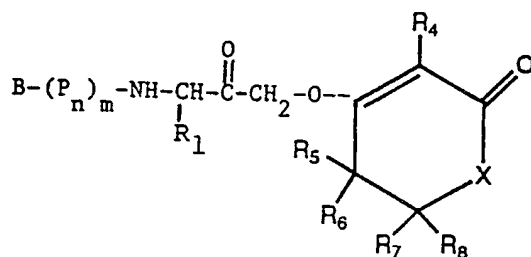
where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 20 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 $R_5$  and  $R_6$  are independently hydrogen, alkyl or  
 25 acyl; and  
 X is N, S, O or  $CH_2$ .

In another embodiment, the cysteine protease inhibitor is of the formula:



- where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 5 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 10 m is 0 or a positive integer;  
 $R_2$  and  $R_3$  are independently H or an alkyl or  
 alkenyl group; and  
 X is N, S, O or  $CH_2$ .

In another embodiment, the cysteine protease inhibitors  
 15 are of the formula:

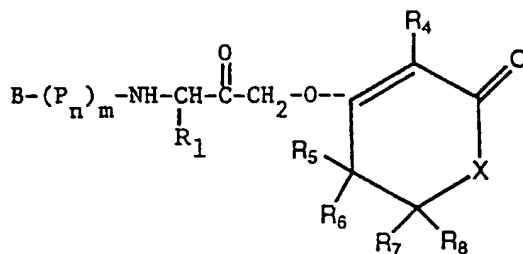




-15-

where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 5 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxy, acyl, hydrogen, alkyl  
 10 or phenyl;  
 $R_5$ ,  $R_6$ ,  $R_7$  and  $R_8$  are independently  
 hydrogen, alkyl, acyl, phenyl, halo, hydroxyl, oxy  
 or alkoxy; and  
 X is N, S, O or  $CH_2$ .

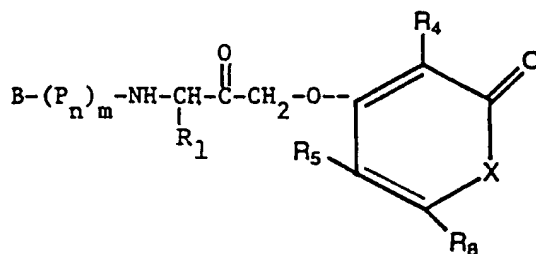
15 In another embodiment, the cysteine protease inhibitors  
 are of the formula:



-16-

- where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 5 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxy, acyl, hydrogen, alkyl  
 10 or phenyl;  
 $R_5$  and  $R_6$  may be attached to  $R_7$  and  $R_8$  to  
 form a ring that is either saturated or unsaturated  
 or aromatic; and  
 X is N, S, O or  $CH_2$ .

- 15 In another embodiment, the cysteine protease inhibitors  
 are of the formula:



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where B is H or an N-terminal blocking group;  
each  $P_n$  is an amino acid residue, or is a  
heterocyclic replacement of the amino acid wherein  
the heterocycle is a piperazine, a  
5 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
pyridone, a carbolinone, a quinazoline, a pyrimidone  
or the like;  
m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxy, acyl, hydrogen, alkyl  
10 or phenyl;  
 $R_5$  and  $R_8$  are independently hydrogen, alkyl,  
acyl, phenyl, halo, hydroxyl, oxy or alkoxy, or  $R_5$   
is attached to  $R_8$  to form a homocyclic or  
heterocyclic ring that is either saturated or  
15 unsaturated or aromatic; and  
X is N, S, O or  $CH_2$ .

One object of the present invention is to provide  
improved cysteine protease inhibitors with improved  
solubility and toxicity profiles.

20 A further object of the present invention is to provide a  
class of cysteine protease inhibitors which are particularly  
effective for in vivo applications.

Further objects and advantages of the present invention  
will be apparent from the following description.

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DESCRIPTION OF THE PREFERRED EMBODIMENT

For the purposes of promoting an understanding of the principles of the invention, reference will now be made to preferred embodiments and specific language will be used to describe the same. It will nevertheless be understood that no limitation of the scope of the invention is thereby intended, such alterations and further modifications of the invention, and such further applications of the principles of the invention as illustrated herein, being contemplated as would normally occur to one skilled in the art to which the invention relates.

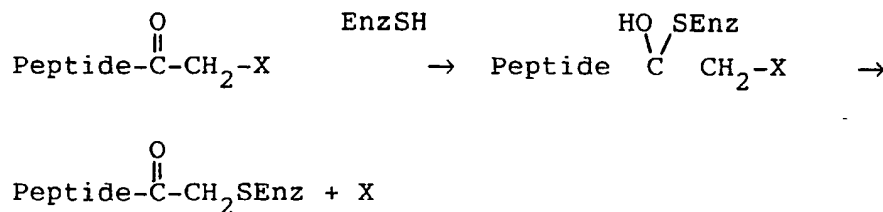
As indicated above, the present invention relates to cysteine protease inhibitors which contain 1,3-dicarbonyl enolether leaving groups. In one aspect of the invention, a group of cysteine protease inhibitors which have been shown to be particularly effective for in vivo applications is disclosed.

The cysteine protease inhibitors described herein function as the sum of two portions. The first portion defines the specificity of a particular inhibitor to an enzyme by the spacial, hydrophobic or hydrophilic and ionic interactions of a particular composition that either imitates or improves upon the nature of the enzyme's natural substrate. The second portion is a trap that covalently binds the enzyme in a two-step mechanism: the first step involves the nucleophilic attack of the enzyme thiolate on the carbonyl of the inhibitor to form a hemithioketal. It is then energetically favorable for this intermediate to undergo a 1,2 migration of the thiolate in which an enolate (or enol form) of a 1,3-dicarbonyl is released. The enzyme has now become irreversibly bonded to the inhibitor. With the inhibitors of the present invention the leaving group is the enol form of a 1,3-dicarbonyl.

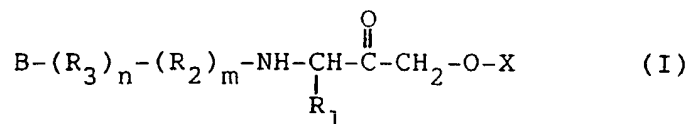
Accordingly, the cysteine proteinase inhibitors of the

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present invention are preferably constructed with an activated carbonyl which bears a suitable  $\alpha$ -leaving group which is fused to a programmed peptide sequence that specifically directs the inhibitor to the active site of the targeted enzyme. (For example, Z-Phe-PheCHN<sub>2</sub> preferentially inhibits cathepsin L over cathepsin B.) Once inside the active site, this inhibitor carbonyl is attacked by a cysteine thiolate anion to give the resulting hemiacetal form. If the  $\alpha$ -leaving group then breaks off, the bond between enzyme and inhibitor becomes permanent and the enzyme is irreversibly inactivated. The selectivity of the inhibitor for a particular enzyme depends not only on the "lock and key" fit of the peptide portion, but also on the reactivity of the bond binding the leaving group to the rest of the inhibitor. It is very important that the leaving group must be reactive to the intramolecular displacement via a 1,2-migration of sulfur in the breakdown of the hemithioacetal intermediate. The mechanism of protease inhibition is shown below in FIG. 1.



The preferred inhibitors of the present invention can be described generally by the formula:



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where

B is H or an N-terminal blocking group;

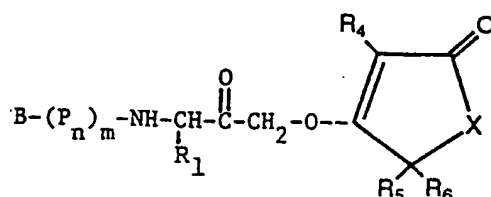
$R_{1-3}$  are the amino acid side chains of the  $P_{1-3}$  amino acids, respectively;

5  $n$  is 0 or 1;

$m$  is 0 or 1; and

X is a five- or six-membered ring portion of the  $\beta$ -dicarbonyl enol ether leaving group, as further defined below.

10 In one embodiment, the compositions of the present invention are cathepsin or calpain inhibitors of the formula:



where B is H or an N-terminal blocking group;

15  $R_1$  is the amino acid side chain of the  $P_1$  amino acid residue; wherein the  $P_1$  amino acid is not Asp; each  $P_n$  is an amino acid residue, or is a heterocyclic replacement of the amino acid wherein the heterocycle is a piperazine, a decahydroisoquinoline, a pyrrolinone, a pyridine, a carbolinone, a quinazoline, a pyrimidone or the like;

20  $m$  is 0 or a positive integer;

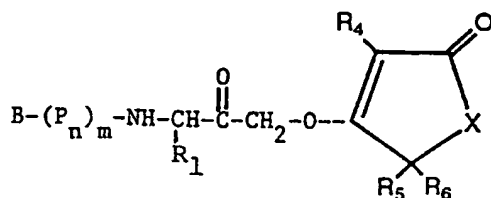
$R_4$  is a hydroxyl, alkoxyl, acyl, hydrogen, alkyl or phenyl;

25  $R_5$  and  $R_6$  are jointly a carboxyl group or a double bond terminating in an alkyl or an aryl group, or are independently acyl, aryl or heteroaryl if  $R_4$  is hydrogen, alkyl or phenyl, or are independently acyl, alkyl, hydrogen, aryl or heteroaryl otherwise; and

X is N, S, O or  $\text{CH}_2$ .

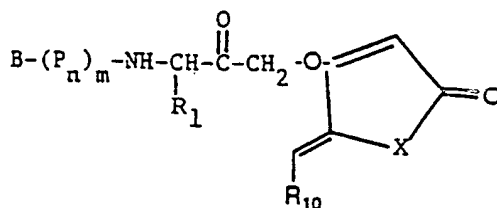
-21-

In another embodiment, the compositions of the present invention are ICE inhibitors of the formula:



where B is H or an N-terminal blocking group;  
 $R_1$  is the Asp amino acid side chain;  
 5 each  $P_n$  is an amino acid residue, or is a heterocyclic replacement of the amino acid wherein the heterocycle is a piperazine, a decahydroisoquinoline, a pyrrolinone, a pyridine, a carbolinone, a quinazoline, a pyrimidone or the like;  
 10 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxy, acyl, hydrogen, alkyl or phenyl;  
 $R_5$  and  $R_6$  are jointly a carboxyl group or a double bond terminating in an alkyl or an aryl  
 15 group, or are independently acyl, aryl or heteroaryl if  $R_4$  is hydrogen, alkyl or phenyl, or are independently acyl, alkyl, hydrogen, aryl or heteroaryl otherwise; and  
 X is N, S, O or  $CH_2$ .

20 In another embodiment, the cysteine protease inhibitor is of the formula:

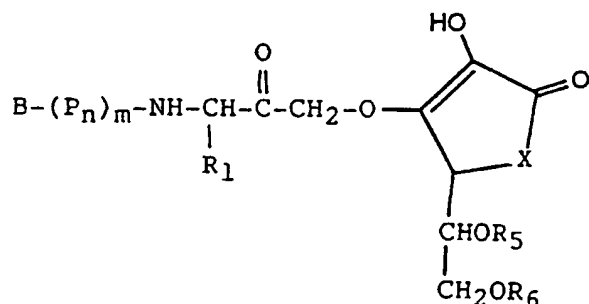


-22-

where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 5 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 carbolinone, a quinazoline, a pyrimidone or the like;  
 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxyl, acyl, hydrogen, alkyl  
 or phenyl;  
 10  $R_{10}$  is H or an optionally substituted alkyl, aryl,  
 heteroaryl, or the residue of a sugar; and  
 X is N, S, O or  $CH_2$ .

In this embodiment X is most preferably oxygen.

In another embodiment, the cysteine protease inhibitor is  
 15 of the formula:



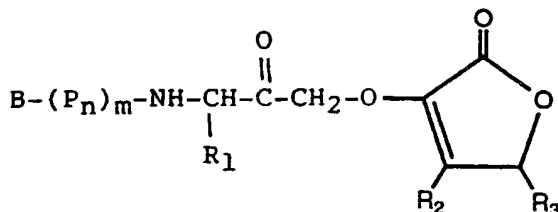
where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 20 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 carbolinone, a quinazoline, a pyrimidone or the like;  
 m is 0 or a positive integer;  
 $R_5$  and  $R_6$  are independently hydrogen, alkyl or  
 acyl; and  
 25 X is N, S, O or  $CH_2$ .



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Most preferably,  $R_5$  and  $R_6$  are each hydrogen. In one alternative embodiment the H on the hydroxyl group of the heterocyclic leaving group may be replaced by an alkyl or alkenyl group.

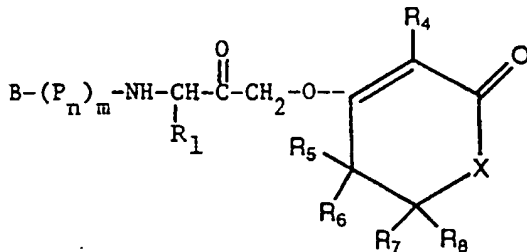
- 5 In another embodiment, the cysteine protease inhibitor is of the formula:



- where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 10 the heterocycle is a piperazine, a  
 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 carbolinone, a quinazoline, a pyrimidone or the like;  
 m is 0 or a positive integer;  
 $R_2$  and  $R_3$  are independently H or an alkyl or  
 15 alkenyl group; and  
 X is N, S, O or  $CH_2$ .

Most preferably  $R_5$  is  $CH_3$  and  $R_6$  is  $C_2H_5$  as shown in compound A2, *infra*.

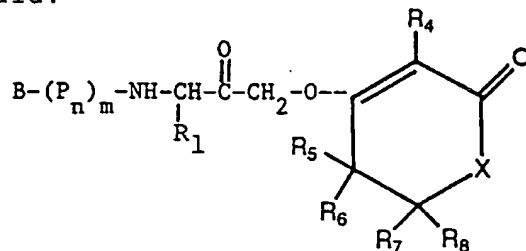
- In another embodiment, the cysteine protease inhibitors  
 20 are of the formula:



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where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 5 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 carbolinone, a quinazoline, a pyrimidone or the like;  
 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxyl, acyl, hydrogen, alkyl  
 or phenyl;  
 10  $R_5$ ,  $R_6$ ,  $R_7$  and  $R_8$  are independently  
 hydrogen, alkyl, acyl, phenyl, halo, hydroxyl, oxy  
 or alkoxy; and  
 X is N, S, O or  $CH_2$ .

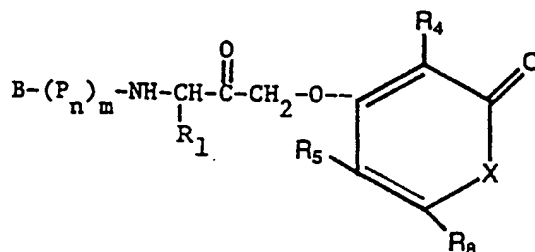
In another embodiment, the cysteine protease inhibitors  
 15 are of the formula:



where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 20 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 carbolinone, a quinazoline, a pyrimidone or the like;  
 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxyl, acyl, hydrogen, alkyl  
 or phenyl;  
 25  $R_5$  and  $R_6$  may be attached to  $R_7$  and  $R_8$  to  
 form a ring that is either saturated or unsaturated  
 or aromatic; and  
 X is N, S, O or  $CH_2$ .

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In another embodiment, the cysteine protease inhibitors are of the formula:



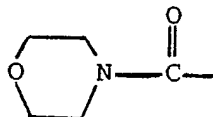
- where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 5 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 carbolinone, a quinazoline, a pyrimidone or the like;  
 m is 0 or a positive integer;  
 10  $R_4$  is a hydroxyl, alkoxy, acyl, hydrogen, alkyl  
 or phenyl;  
 $R_5$  and  $R_8$  are independently hydrogen, alkyl,  
 acyl, phenyl, halo, hydroxyl, oxy or alkoxy, or  $R_5$   
 is attached to  $R_8$  to form a homocyclic or  
 15 hererocyclic ring that is either saturated or  
 unsaturated or aromatic; and  
 X is N, S, O or  $CH_2$ .

As to the amino acid blocking group B for the N-terminal  
 amino acid nitrogen, many suitable peptide end-blocking  
 20 groups are known in the art. For example the end-blocking  
 groups identified in E. Gross and J. Meienhofer (eds.), The  
 Peptides, Vol. 3 are generally suitable for use in the  
 present invention. Preferred blocking groups include  
 N-morpholine carbonyl and derivatives of propionic acid  
 25 derivatives that have intrinsic analgesic or  
 anti-inflammatory action. Examples of blocking groups having

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intrinsic analgesic or anti-inflammatory action may be found in Gilman, Goodman, Gilman, The Pharmacological Basis of Therapeutics, Sixth Ed. MacMillan, Chapter 29. As defined herein, the peptide end-blocking group is attached to either  
5 an amino acid or a peptide chain.

One particularly effective blocking group is the 4-morpholinylcarbonyl ("Mu") blocking group shown below:



4-morpholinylcarbonyl

Other useful blocking groups include the morphine sulfonyl group and related groups as reported by Doherty et al. in "Design and Synthesis of Potent, Selective and Orally  
10 Active Fluorine-Containing Renin Inhibitors," 35 J. Med. Chem. 2. An appropriate blocking group for a particular inhibitor may be selected by persons skilled in the art without undue experimentation.

As is conventional in the art, and as used herein, amino acid residues are generally designated as  $P_1$ ,  $P_2$ , etc., wherein  $P_1$  refers to the amino acid residue nearest the leaving group,  $P_2$  refers to the amino acid residue next to  $P_1$  and nearer the blocking group, etc. In dipeptide  
20 inhibitors therefore,  $P_2$  is the amino acid residue nearest the blocking group. In this disclosure the chain of amino acid residues is frequently written  $(P_n)_m$  with each  $P_n$  being an amino acid residue and  $m$  being zero or a positive integer. Each  $P_n$  may, of course, be a different amino acid residue. Preferably,  $m$  is less than or equal to four. Most  
25 preferably,  $m$  is two.

As suggested above, any of the amino acid residues may be replaced by a heterocyclic replacement. Preferably the heterocycle is a piperazine, a decahydroisoquinoline, a  
30 pyrrolinone, a pyridine, a carbolinone, a quinazoline, a

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pyrimidone or the like. Persons skilled in the art may select an appropriate heterocycle in a manner similar to that in which appropriate amino acid residues are selected. As used herein therefore, terminology such as "the peptide  
5 portion" refers as well to the corresponding portion when heterocycles replace any or all of the amino acids.

The peptide portion of the inhibitor includes any peptide appropriate for targetting a desired cysteine protease. In particular, the side chain on the  $P_1$  amino acid is selected  
10 according to the enzyme being targetted. For cathepsin B or L, this might include side chains such that the linked  $P_1$  amino acid is a member of the group consisting of alanyl (Ala), arginyl (Arg), glutamic acid (Glu), histidyl (His), homophenylalanyl (HPhe), phenylalanyl (Phe), ornithyl (Orn),  
15 seryl (Ser) and threonyl (Thr), and optionally substituted analogues thereof such as thiazoles and amino thiazoles. Preferably the side chain on the  $P_2$  amino acid is selected so that the linked  $P_2$  amino acid is a member of the group consisting of phenylalanyl (Phe), leucyl (Leu), tyrosyl (Tyr)  
20 and valyl (Val) amino acid residues and substituted analogues thereof, particularly including Tyr(OMe).

More specifically regarding the selection of side chains, the cathepsins and the calpains share great cross reactivities with many inhibitors of structures shown above,  
25 although Cathepsin B responds most strongly to basic side chains at  $P_1$  (although reacting to several), while Cathepsin L is more susceptible to neutral side chains at  $P_1$ . Both Cathepsin B and Cathepsin L require neutral side chains at  $P_2$ . Cathepsins H and C prefer to attach to  
30 unblocked peptides, with Cathepsin H favoring a single peptide, Cathepsin C a dipeptide and the calpains susceptible to neutral side chains. The cathepsins, as a general rule, are more reactive than the calpains. Interestingly, neither of these two enzyme types is inhibited when Asp occupies the  
35  $P_1$  position. In contrast, the interleukin-1 $\beta$ -converting

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enzyme (ICE) is unaffected by all these inhibitors unless Asp is at the  $P_1$  position. This fundamental difference between the ICE enzyme and its inhibitors on the one hand, and the other cysteine enzymes and their inhibitors on the other, is well documented in the literature.

When an aspartyl side chain is present in inhibitors based on an activated ketone an unnatural event occurs - the free acid in the side chain attacks the ketone (whose counterpart in the natural substrate is an unreactive amide carbonyl) and a thermodynamically favored hemiketal results. Such hemiketals may be transition state mimics that are also known to play a role in protease inactivation. The inhibition of the ICE enzyme can now follow either of two paths: hemiacetal exchange or thiolate attack on unmasked ketone, thus leading to some confusion in the detailing of the mechanism of inhibition. The problem is eliminated by esterification of the side chain.

One optimum peptide sequence for ICE inhibitors is known to be: B - Tyr - Val - Ala - Asp - Trap; where B is the blocking group and the "trap" is the activated ketone or aldehyde (reversible inhibitor). Dolle has shown that this sequence can be reduced to Val - Ala - Asp - and even Asp alone is inhibitory. Dolle et al.,  $P_1$  Aspartate - Based Peptide  $\alpha$ -((2,6-Dichlorobenzoxymethyl)ketones as Potent Time-Dependent Inhibitors of Interleukin-1 $\beta$ -Converting Enzyme, 37 J. Med. Chem. 563.

The leaving groups of the present invention share certain features to assure the low toxicity and good solubility of the inhibitors. In particular, the leaving groups of the present invention: (1) immitate or improve upon the cleaved peptide portion of the proteases natural substrate; (2) activate the carbonyl of the inhibitor to selectively react with the thiolate of a cysteine protease; (3) are non-toxic and non- cleavable by non-cysteine proteases and esterases; and (4) are very water soluble and enable the use of more

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amino acids than prior art leaving groups allow.

As indicated, the inhibitors of the present invention immitate or improve upon the cleaved peptide portion of the proteases natural substrate. The natural substrate leaving  
5 group is the sum of planar (or nearly so) amide bonds fused by tertiary substituted chiral carbon atoms. In applicant's prior application (Ser. No. 08/164,031) it was disclosed that oxygen fused to a heterocyclic ring could imitate the planar features of the natural substrate leaving groups, and also  
10 that a heterocycle unit provides the diversity needed to imitate the different electronic and spacial specificity requirements of different individual enzymes. It was also demonstrated that the degree of aromaticity of the ring was not a requirement for efficacy.

15 The inhibitors of the present invention activate the carbonyl of the inhibitor to selectively react with the thiolate of a cysteine protease. This concept was not appreciated until the demonstration of the success of peptidyl fluoroketones by Rasnick et al. in U.S. Patent  
20 4,518,528. The best replication of the chemistry ascribed to the most electronegative atom is to replace fluorine with the second most electronegative atom (oxygen) deshielded further by the attachment of an atom with electron withdrawing double bond character. The inhibitors of this invention maximize  
25 this premise by electronically coupling the anion of the leaving group to the electo positive center of a carbonyl carbon. By using a hydrocarbon structure devoid of halogens we eliminate the toxicity associated with peptidyl fluoroketones, trifluoromethyl substitutions, and halogenated  
30 hydrocarbons which are common to other inhibitors in the art.

The inhibitors of the present invention are non-toxic and non-cleavable by non-cysteine proteases and esterases. In an attempt to minimize dipole moments, 1,3-dicarbonyls form very stable enols, and as a result the  $\alpha$ -ketoethers prepared in  
35 this invention show outstanding stability and oral efficacy.

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On the other hand, 1,3-dicarbonyls are readily eliminated through the Krebs Cycle and therefore pose less of a toxicity potential than nitrogen aromatic heterocycles and other aromatics that require liver oxidative clearance.

5       The inhibitors of the present invention are preferably very water soluble and enable the use of more amino acids than current art leaving groups in the peptide construction of the inhibitor. One generalization that can be made about the state of the art inhibitors is that the leaving group is  
10       of high molecular weight (as a dichlorophenol) which reduces the overall water solubility and oral efficacy of the peptide inhibitor. The smaller more polar oxyheterocycles of this invention actually increase the water solubility of the peptide inhibitor as shown in Example 8 where the leaving  
15       group derived from tetronic acid increases the solubility of the peptide inhibitor almost two fold over that of the peptide portion alone (approximated by the fluoroderivative). Further additions of hydroxyl groups to the parent tetronic nucleus further enhances water solubility  
20       and leaving groups such as those derived from ascorbic acid becomes most preferable.

Reference will now be made to specific examples for making and using the cysteine protease inhibitors of the present invention. It is to be understood that the examples  
25       are provided to more completely describe preferred embodiments, and that no limitation to the scope of the invention is intended thereby.

#### EXAMPLE 1

30       N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(4-oxy-5-phenyl-4-cyclopentene-1,3-dione) methyl ketone.



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N-morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine bromomethyl ketone (100 mg, 0.194 mmol), potassium fluoride (45 mg, 0.775 mmol), and 4-hydroxy-5-phenyl-4-cyclopentene-1,3-dione was placed in a 20 cm test tube equipped with a stirring bar and placed under an argon atmosphere. Next 3 ml of dry DMF was syringed into the reaction which was allowed to stir at room temperature until TLC (silica gel, CHCl<sub>3</sub>/isopropanol:95/5) showed total loss of starting material. The reaction was then passed through a short plug of silica gel (ethyl acetate) and the solvent was removed in vacuo. The resulting material was purified by size exclusion chromatography (LH 20, methanol) and precipitated in ether to give a yellow powder after filtration. (m.p. = 155-157°C, IC<sub>50</sub> Cathepsin B, 94 nM.)

#### EXAMPLE 2

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanyl- $\alpha$ -(4-ascorbityl) methyl ketone.

N-morpholinecarbonyl-L-phenylalanine bromomethylketone (495 mg, 1 mmol), sodium ascorbate (380 mg, 2 equivalents), and potassium fluoride (116 mg, 2 equivalents) was placed in a 50 mL round bottom flask under an atmosphere of argon. Next, 5 ml of dry DMF was syringed in and the reaction was allowed to stir at room temperature overnight. The next day the reaction was filtered through celite and the solvents were removed in vacuo.

The residue was dissolved in chloroform and the resulting solution was diluted with an equal volume of methylene chloride to precipitate the unreacted sodium ascorbate. After filtration the solvent was removed in vacuo and the residue purified by size exclusion chromatography to give a

-32-

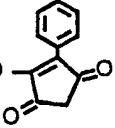
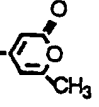
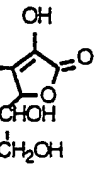
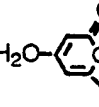
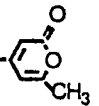
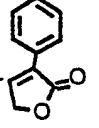
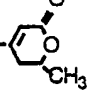
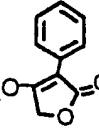
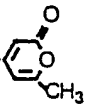
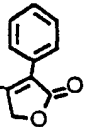
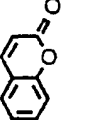

white solid, mp 105-110°C. IC<sub>50</sub> Cathepsin B, 141 nM.

In a similar manner the following compounds were prepared:

- 5 N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanyl- $\alpha$ -(4-oxy-6-methyl-2-pyrone) methyl ketone (m.p. 94-98°C);  
N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanyl- $\alpha$ -(4-oxy-5,6-dihydro-6-methyl-2H-pyran-2-one) methyl ketone (m.p. 74-78°C); N-Morpholinecarbonyl-L-Leucyl-L-homophenylalanyl- $\alpha$ -(4-oxy-(6-methyl-2-pyrone) methyl ketone (m.p. 70-75°C);  
10 N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanyl- $\alpha$ -(4-oxy-coumarin) methyl ketone (m.p. 115-119°C);  
N-Morpholinecarbonyl-L-phenylalanyl-L-lysyl-(4-oxy-(6-methyl-2-pyrone) methyl ketone (m.p. 151-155°C); N-Morpholinecarbonyl-L-tryosyl (O-methyl)-L-lysyl-(4-oxy-(6-methyl-2-pyrone) methyl ketone (m.p. 140-142°C);  
15 N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanyl- $\alpha$ -(4-oxy-3-phenyl-dihydrofuran-2-one) methyl ketone (m.p. 114-116°C);  
N-Morpholinecarbonyl-L-tryosyl (O-methyl)-L-lysyl-(4-oxy-3-phenyl-dihydrofuran-2-one) methyl ketone (m.p. 140-142°C); N-Morpholinecarbonyl-L-phenylalanyl-L-lysyl- $\alpha$ -(4-oxy-3-phenyl-dihydrofuran-2-one) methyl ketone (m.p. 140-145°C); N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanyl- $\alpha$ -(4-oxy-dihydrofuran-2-one) methyl ketone  
20 (m.p. 155-157°C).  
25

Structural formulas and in vitro activities (against Cathepsin B) for these compounds are set forth below:

*In Vitro* Activity Against Purified Cathepsin B

Compound	IC <sub>50</sub> Cat B	Compound	IC <sub>50</sub> Cat B
1. Mu Phe Hphe CH <sub>2</sub> O- 	94 nM	7. Mu Phe Lys CH <sub>2</sub> O- 	25 nM
2. Mu Phe Hphe CH <sub>2</sub> O- 	141 nM	8. Mu Tyr(OMe) Lys CH <sub>2</sub> O- 	83 nM
3. Mu Phe Hphe CH <sub>2</sub> O- 	112 nM	9. Mu Phe Hphe CH <sub>2</sub> O- 	36 nM
4. Mu Phe Hphe CH <sub>2</sub> O- 	567 nM	10. Mu Tyr(OMe) Lys CH <sub>2</sub> O- 	8 nM
5. Mu Leu Hphe CH <sub>2</sub> O- 	400 nM	11. Mu Phe Lys CH <sub>2</sub> O- 	5 nM
6. Mu Phe Hphe CH <sub>2</sub> O- 	45 nM	12. Mu Phe Hphe CH <sub>2</sub> O- 	10 nM

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EXAMPLE 3Protocol for the in vitro evaluation  
of inhibitors with cathepsin B.

Enzyme: Cathepsin B, purified from human liver, is from  
5 Enzyme Systems Products (Dublin, CA). The activity is 50 mU  
per ml at 30°C, in 52 mM sodium phosphate, pH 6.2, 31 mM DTT,  
2.1 mM EDTA, with 0.2 mM Z-Arg-Arg-7-amino-4-trifluoromethyl-  
coumarin as a substrate. Specific activity is 8330 mU per mg  
protein. (1 mU = 1 nmol per min.)

10 Substrate Boc-Leu-Arg-Arg-7-amino-4-trifluoromethyl-  
coumarin-2HBr is from Enzyme Systems Products (Dublin, CA)  
and is known to be a specific substrate for cathepsin B. A  
20 mM solution is made in DMF and stored at 20°C.

Candidate inhibitors are dissolved in DMF and diluted to  
15 20 mM and stored at 20°C. Dilutions are made in assay buffer.

The percent inhibition and the inhibitor concentration at  
which the enzyme is 50% inhibited ( $IC_{50}$ ) are determined as  
follows. Five  $\mu$ l of assay buffer (50 mM potassium phosphate  
pH 6.2, 2 mM EDTA, 5mM DTT) on ice for 30 min. The  
20 inhibition is initiated by the addition of 5 ml of 200 mM, 20  
mM, and 2 mM inhibitor each to the 480  $\mu$ l aliquots. The 485  
 $\mu$ l aliquot with enzyme is used as a control and thus

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receives no inhibitor. The enzyme/inhibitor mixtures are incubated 10 min. on ice and assayed for cathepsin B activity as follows: Cathepsin B assay: To 490  $\mu$ l of pre-incubated inhibitor/enzyme mixtures in assay buffer in 0.5 ml cuvette  
5 at 37° is added 10  $\mu$ l of the substrate. Final inhibitor concentrations become 2000 nM, 200 nM, and 20 nM for the 200  $\mu$ M, 20  $\mu$ M and 2  $\mu$ M stock concentrations, respectively. Activity is followed by release of free AFC over 5 min. The change in fluorescence is (fluorescence units at t=6) -  
10 (fluorescence units at t=1) with a Perkin-Elmer LS-5B spectrofluorometer (ex = 400 nm, em = 505 nm). The percent inhibition is determined by comparing the change in fluorescence units of the three sample concentrations of inhibited enzyme to the change in fluorescence units of the  
15 control enzyme. The percent inhibition is calculated as:  
100 - (fl. units of sample/fl. units of control x 100).  
The IC<sub>50</sub> is ascertained by plotting percent inhibition vs. inhibitor concentration on the log scale. The IC<sub>50</sub> is the concentration of the inhibitor (nM) at which the enzyme  
20 is inhibited by 50%.  
IC<sub>50</sub> values for preferred inhibitors are shown on the table of structural formulas supra.

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EXAMPLE 4

Table 1. Malarial Cysteine Protease Inhibition: IC<sub>50</sub> Concentrations<sup>1</sup>

	<u>New Inhibitor</u>	<u>P. falciparum</u>	<u>P. vinckei</u>
5	6.	5-10 nM	5-10 nM
	7.	~50 nM	~100 nM
	9.	300-500 pM	<1nM
	11.	5-10 nM	~10nM
	12.	~10 nM	~ 10nM
10	Proteolytic activity assays. Gelatin-substrate PAGE is performed described in Rosenthal, McKerrow, Rasnick, and Leech, Plasmodium falciparum: Inhibitors of Lysosomal Cysteine Proteinases Inhibit a Trophozoite Proteinase and Block Parasite Development, 35 <u>Mol. Biochem. Parasitol.</u> 177-184 (1989). In brief, this technique involves		
15	electrophoresis of nonreduced proteins on a gelatin-containing gel, removal of SDS from the gel by washing with 2.5% Triton-100, overnight incubation (0.1 M sodium acetate, 10 mM dithioerythritol (pH 6.0, 37°C) of the gel to allow hydrosis		
20	of the gelatin by renatured proteinases, and staining with Coomassie blue. Proteinases are identified as clear bands in the blue staining gel. To evaluate the effects of proteinase inhibitors, the inhibitors are incubated with parasite		
25	extracts (1 hr, room temperature) before samples are mixed with the electrophoresis sample buffer, and they are		

<sup>1</sup> The IC<sub>50</sub> is the concentration of the inhibitor (nM) at which the enzyme is inhibited 50% within 6 minutes in our standard in vitro assay (see Example 3).

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included in the overnight gel incubation buffer. Proteolytic activity was also measure with two other substrates: (a) fluorogenic peptidesubstrates containing the 7-amino-4-methyl- coumarin (AMC) detecting group (Enzyme Systems Products, Dubic, CA) and (b) [<sup>14</sup>C]-methemoglobin (Dupont New England Nuclear, Wilmington, DE), both as described in Rosenthal, McKerrow, Aikawa, Nagasawa, and Leech, A Malarial Cysteine Proteinase is Necessary for Hemoglobin Degradation by Plasmodium Falciparum. 82 J. Clin. Invest. 1560-66.(1988).

In another test of malarial inhibition by the cysteine protease inhibitors of the present invention compound A2; infra, was particularly effective.

#### EXAMPLE 5

Table 3. Inhibition of T. cruzi in Infected Cells with New Inhibitors.

<u>Compound</u>	<u>Survival Time</u>	
	<u>Cell line J774</u>	<u>Cell line BHK</u>
Control	4 Days	5 Days
Mu Phe HPheCH <sub>2</sub> F	16 Days plus	16 Days plus
6.	16 Days plus	16 Days plus
9.	4 Days	6 Days

Survival time is measured in days before the cell monolayer is destroyed by the infection. Irradiated BHK and J774 cells (six well plates) were infected with T. cruzi and simultaneously treated with 20 μM (3 ml total volume) with daily change of culture medium+plus inhibitor).

Cultivation and preparation of T. cruzi. Cloned and uncloned populations were derived from the strains Brasil and CA-I and are cryopreserved in liquid nitrogen. Axenically cultured epimastigotes are maintained in exponential growth phase by weekly passage in Brain Heart Infusion-Tryptose medium (BHT media as given in Cazzulo, Cazzulo, Martinez, and Cazzulo, Some Kinetic Properties of a Cysteine Proteinase (Cruzipain) from Trypanosoma Cruzi. 33 Mol. Biochem.

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Parasitol. 33-42 (1990)), supplemented with 20 µg/ml and 10% (v/v) heat inactivated fetal calf serum (FCS). Different host cell lines (J774 mouse macrophage, BHK, etc.) are cultured with RPMI-1640 supplemented with 5% FCS at 37° in a humidified atmosphere containing 5% CO<sub>2</sub>. Trypomastigotes liberated from the host cells are used to infect new cultures for serial maintenance of the parasite in cell culture. The protocols used for the in vitro assays of cysteine protease inhibitors are essentially as described by Harth, Andrews, Mills, Engel, Smith, and McKerrow, Peptide-Fluoromethyl Ketones Arrest Intracellular Replication and Intercellular Transmission of Trypanosoma Cruzi. 58 Mol. Biochem. Parasitol. 17-24 (1993), with the exception that in some experiments, the host cells are irradiated (2400 RADs) before infection to prevent them from dividing.

#### EXAMPLE 6

##### In vitro inhibition of Pneumocystis carinii.

When the following compounds were tested in a Pneumocystis carinii culture system with human embryonic lung fibroblast monolayers, the organism proliferation was inhibited as demonstrated below. The percent of inhibition is calculated as 100-(the number of P. carinii trophozoites in a treated cell culture/number of trophosites of the control) X 100.

<u>Compound</u>	<u>Percent Inhibition</u>			
	<u>Day 1</u>	<u>Day 3</u>	<u>Day 5</u>	<u>Day 7</u>
Control	0	0	0	0
3	23	44	39	13
9	17	67	64	65

Methods. The drugs dissolved in dimethyl sulfoxide were diluted to concentrations of 10 µg/ml for compound 3 and 10



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μM/ml for compound 9 in minimum essential medium used for  
 the culture of human embryonic lung fibroblasts. The final  
 maximum dimethyl sulfoxide concentration was 0.1%, a  
 concentration of dimethyl sulfoxide that did not affect P.  
 5 carinii proliferation when it is used alone and that gave P.  
carinii growth curves comparable to those of organisms in  
 untreated control wells. Cell cultures in 24-well plates  
 were inoculated with P. carinii trophozoites (final  
 concentration, about  $7 \times 10^5$  per ml) obtained from infected  
 10 rat lungs. Each culture plate contained untreated and  
 treated wells. Plates were incubated at 35°C in a gas  
 mixture of 5% O<sub>2</sub>, 10% CO<sub>2</sub>, 85% N<sub>2</sub> for up to 7 days.  
 Plates were sampled on days 1, 3, 5 and 7 by removal of 10  
 μl amounts after agitation of the cultures. The samples  
 15 were placed on slides in 1-cm<sup>2</sup> areas, fixed in methanol and  
 stained with Giemsa stain; and then they were examined  
 microscopically as unknowns by two individuals. For each  
 parameter there were four wells, making eight values for each  
 parameter. Standard errors range from 3-13%. Cultures were  
 20 spiked with fresh drugs on days 2, 4 and 6.

#### EXAMPLE 7

The in vivo inhibition of cathepsin B in rat liver

Compound	Time Post Dose (Hours)				
	<u>1.5</u>	<u>3</u>	<u>6</u>	<u>12</u>	<u>24</u>
(via stomach tube).					
25 6	48	29	N/A	30	0
12		53	32		
(1600 nM dose)					
12		35	31		
30 (800 nM dose)					
(via injection, IP)					
12		79	70		
(1600 nM dose)					
12		55	70		
35 (800 nM dose)					

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Protocol for the in vivo Evaluation of Inhibitors.

Female Sprague Dawley rats (150-200g each) are purchased from Simonson, Gilroy, CA. After 1 week of acclimation in-house, the animals (usually four per group) are dosed by the  
5 selected route of administration. Test compounds are dissolved in ethanol and diluted to the appropriate concentration with water. In control studies, animals are dosed only with ethanol water vehicle.

Tissue Homogenate Preparation. At the appropriate time  
10 post dose, the treated animals are anesthetized with ether, decapitated and exsanguinated. The tissues of interest are removed, quickly frozen in liquid nitrogen and then are stored at -70°C until processing. All subsequent manipulations of the tissue samples are carried out at 4°C.  
15 Liver and skeletal muscle are pulverized while still frozen and then homogenized, while other target tissues are homogenized without prior pulverization. The tissue homogenization, in distilled water or 0.1% Brig-35, are subsequently performed using three 15-s bursts with a 10 N  
20 probe on a Tekmar Tissuemizer set to 75-80% power. The samples are centrifuged at 15000 g for 40 min; they partition into a lipid layer, a lower clarified layer and a solid pellet. The clarified supernatant is carefully aspirated and transferred to clean polypropylene tubes for storage at  
25 -70°C, until the fluorometric assay for enzyme activity can be performed.

Purified Lysosomal Enzyme Preparation. The procedure is based on a report by Bohley et al. (1969) and Barrett and Kirshke (1981). At the appropriate time post dose, the  
30 treated animals are anesthetized with sodium barbital and the livers are perfused in situ with ice-cold saline. The livers are then removed, rinsed with ice-cold saline, blotted and weighed. The animals are sacrificed with ether. All subsequent manipulations of the tissue samples are carried

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out at 4°C. The livers are homogenized in 2 volumes of 0.25M sucrose at 0°C with a 30 ml Wheaton Teflon-on-glass homogenizer, using five full strokes with a motor setting at 55. Following centrifugation at 600 g for 10 minutes, the  
5 supernatant is transferred to clean tubes for centrifugation at 3000 g for 10 minutes. The resulting supernatant is centrifuged for 15 minutes. The lysosomal pellet is washed twice with 0.25 M sucrose, lysed in 2.5 volumes of distilled water using a glass-on-glass homogenizer, and then  
10 centrifuged at 1000 g for 60 minutes. The supernatant is stored at -70°C until fluorimetric assay for enzyme activity is performed.

## EXAMPLE 8

Aqueous solubilities of Morpholinecarbonyl-phenylalanyl-  
15 homophenylalanyl- $\alpha$ -(4-oxy-dihydrofuran-2-one) methyl ketone vs. Morpholine-carbonyl-phenylalanyl-homophenylalanyl-fluoromethylketone.

The aqueous solubilities of the two title compounds at 20°C were determined by using UV spectroscopy and compared to  
20 that of a known standard benzoxycarbonyl-phenylalanyl-alanylfluoromethyl ketone. The aqueous solubility of Mu-Phe-HPhe- $\alpha$ -(4-oxy-dihydrofuran-2-one) methyl ketone 12 was measured to be 0.277 mg/ml. The aqueous solubility of Mu-Phe-HPhe-CH<sub>2</sub>F was measured to be 0.140 mg/ml. The  
25 aqueous solubility of Z-Phe-Ala-CH<sub>2</sub>F was 0.045 mg/ml at 14°C.

Experimental. A saturated solution was prepared by weighing 10 mg of the material and placing it in 5 ml of distilled and deionized water. The solution was capped and  
30 stirred at 20°C unless otherwise noted. A 1 ml aliquot was removed after 24 hours and was filtered through a .45 $\mu$ m filter and diluted 1:50 with distilled and deionized water. Subsequent aliquots were taken and similarly diluted after 48, 72, and 96 hours respectively. The absorbances were

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measured at 247 nm for compound 12 and at 219nm for Mu-Phe-HPhe-CH<sub>2</sub>F and were compared against a series of respective standard solutions run under similar conditions.

## EXAMPLE 9

5                   Synthesis of the cathepsin H inhibitors.

L-Homophenylalanyl- $\alpha$ -(4-oxy-(6-methyl-2-pyrone) methyl ketone. BOC-homophenyl-bromomethylketone (300 mg), potassium fluoride (195 mg), potassium carbonate (233 mg) and 4-hydroxy-6-methyl-2-pyrone (212 mg) was placed in a round  
10 bottom flask under an atmosphere of argon. About one mL of DMF was added and the mixture was stirred at 50°C for 40 min. The reaction was then diluted with ethyl acetate (10X) and passed through a plug of silica gel to remove the salts. The solvents were removed under vacuum. The BOC-protecting  
15 group was removed by dissolving the resulting solid in 3 mL of methylene chloride and adding 3 mL of 4N HCl-dioxane. The reaction was run until only a stationary spot was detected with silica gel TLC (9:1, CHCl<sub>3</sub>:isopropanol). The resulting mixture was then added dropwise to 50 mL of ether  
20 and the precipitated solid was filtered. mp. 177-179°C. IC<sub>50</sub> Cathepsin H: 118 nM.

          In the same manner L-Homophenyl- $\alpha$ -(4-oxy-dihydrofuran-2-one)methyl ketone hydrochloride was synthesized. mp. 128-132°C. IC<sub>50</sub> Cathepsin H: 251 nM.

25           Numerous other cathepsin H inhibitors can be made with the construction of an unblocked amino acid on an  $\alpha$ -oxy heterocycle methyl ketone.

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EXAMPLE 10  
SYNTHESIS OF ICE INHIBITORS

The following example is meant to be illustrative but is not meant to be restrictive to other variations which would involve exchanges of blocking groups, abbreviation or minor alterations in side chain construction, or exchanges with other leaving groups of this invention.

N-Benzoxycarbonyl-valyl-alanyl-aspartyl- $\alpha$ -(4-oxy-(6-methyl-2-pyrone) methyl ketone. Z-Val-AlaOMe: N-Benzoxycarbonyl-valine was dissolved under argon in 300 mL of freshly distilled THF and the resulting solution was cooled in a methanol-ice bath. One equivalent of N-methyl morpholine followed by one equivalent of isobutylchloroformate was added and the reaction was allowed to activate for 20 minutes. Another equivalent of N-methylmorpholine is then added followed by one equivalent of solid alanine methyl ester hydrochloride salt. The reaction was allowed to come slowly to room temperature and stir overnight. The next day the reaction was poured into 200 mL of 1N hydrochloric acid and extracted with ethyl acetate (2 x 150 mL). The combined organic fractions were washed with brine (50 mL), aqueous sodium bicarbonate (100 mL), dried over  $\text{MgSO}_4$ , filtered and the solvents were removed under reduced pressure to give 14 g of a white solid methyl ester. mp. 157-163°C.

Hydrolysis of the methyl ester was effected by dissolving 2.20 g in 35 mL of methanol followed by the addition of 8.2 mL of 1N aqueous sodium hydroxide solution. The reaction was stirred at room temperature for 4 hours. At this time TLC (silica gel/ $\text{CHCl}_3$ /isopropanol) showed that most of this material had been converted to the acid (stationary spot on TLC). The methanol was then removed under reduced pressure and the residue was dissolved in water (100 mL) an additional

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5 mL of sodium hydroxide was added and the water was washed with ethyl acetate (50 ml) and then neutralized with 1N hydrochloric acid, and extracted with 2 x 100 ml of ethyl acetate. The organic layer was dried over  $\text{MgSO}_4$ , filtered and the solvents were evaporated to give a white solid: mp. 170-175°C.

Condensation with aspartyl (O-t-butyl)-O-methyl ester was effected as follows: Z-Val-Ala-OH was dissolved in 300 mL of freshly distilled THF and the resulting solution was cooled in an ice-methanol bath. Next one equivalent of N-methyl morpholine was added followed by one equivalent of isobutylchloroformate and the reaction was allowed to activate for 20 minutes. Another equivalent of N-methyl morpholine was added followed by one equivalent (5g) of HCl-Asp(OtBu)OMe. The mixture was allowed to come slowly to room temperature and stir overnight. The next day the reaction was poured into 200 ml of 1N hydrochloric acid and extracted with ethyl acetate. The organic layer was washed with sodium bicarbonate (aq, 50 mL), brine (50 mL) and the organic layer was dried over  $\text{MgSO}_4$ , filtered, and the solvents were removed under reduced pressure. The residue was crystallized from 50 mL of methylene chloride and 200 mL of ether to give white crystals (4.0 g), mp. 157-163°C.

Hydrolysis to Free Acid: Z-Val-Ala-Asp(otBu)OMe (4.6 g) was dissolved in 30 mL methanol and then 12 mL of 1N sodium hydroxide (aq) was added and the reaction was stirred for 1 hour at room temperature. At the end of this time the methanol was removed under reduced pressure and 50 ml of water plus another 12 mL of 1N sodium hydroxide was added to dissolve the precipitated solid. The resulting water solution was washed with ethyl acetate (50 mL) and then the water fraction was acidified with 1N HCl and the resulting mixture extracted with ethyl acetate, dried over  $\text{MgSO}_4$ , filtered and concentrated to give 3.74 g of Z-Val-Ala-Asp(O-tBu)OH.

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Conversion to the Diazoketone: Z-Val-Ala-Asp(O-tBu)OH was dissolved in 200 mL of freshly distilled THF and a methanol-ice bath was applied. Next one equivalent of N-methyl morpholine followed by one equivalent of isobutyl chloroformate was added and the reaction was allowed to activate for 20 minutes and then the resulting mixture was poured through filter paper into diazomethane/ether made from 6.3 g of Diazald (Aldrich) according to the supplier's directions. The reaction was allowed to stand overnight and then worked up in the following way: The reaction was washed with water (2 x 50 mL), sodium bicarbonate (50 mL), brine (50 mL) and then dried over  $\text{MgSO}_4$ , filtered and the solvents were removed under reduced pressure to give a yellow solid 3.75 g. This residue was then chromatographed in two parts through a 1 x 12 inch silica gel ( $\text{CHCl}_3$ :isopropanol/95:5) column to give two product fractions. The product with the lower  $R_f$  value (0.3) was shown by the absorption at 5.54 ppm in the 100 MHz NMR to be the correct product.

Conversion to the Bromoketone: Z-Val-Ala-Asp(OtBu)CHN<sub>2</sub> was dissolved in 25 mL ether and 25 mL THF and a methanol-ice bath was applied. Next 0.1 mL HBr/acetic acid (30%) diluted to 10 mL with ether:THF (1:1) was added dropwise. The yellow solution becomes clear and when no more color remains the reaction is poured into an equal volume of brine, the organic layers are separated and the water fraction is washed with an additional 50 mL THF:ether. The organic fraction was then washed with 50 mL of aqueous sodium bicarbonate, 50 mL brine, dried over  $\text{MgSO}_4$ , filtered and concentrated to give a white solid: mp. 150-151°C.

Conversion to the  $\alpha$ -(4-oxy-(6-methyl-2-pyrone) methyl ketone: Z-Val-Ala-Asp(otBu)CH<sub>2</sub>Br (131 mg), 4-hydroxy-6-methyl-pyrone (58 mg, 2 equivalents), potassium fluoride (53 mg), and 1.5 mL of DMF was stirred at room temperature for two hours at which time TLC (silica gel,  $\text{CH}_3\text{Cl}$ /isopropanol:97/3) showed loss of starting material

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and development of product. The reaction was then run through a plug of silica gel ( $\text{CHCl}_3$ /isopropanol/9:1) and the solvents were removed under reduced pressure. Most of the excess pyrone starting material was precipitated from isopropyl ether:  $\text{CH}_2\text{Cl}_2$  and the  
5 Z-Val-Ala-Asp(OtBu)- $\alpha$ -(4-oxy-6-methyl-pyrone) methyl ketone was isolated from the resulting mother liquor by removal of the solvent and size exclusion chromatography: NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  0.9 (dd, 6), 1.4 (broad s+d, 12), 2.1  
10 (s, 3), 3.5 (s, 2), 5.1 (s, 2), 7.3 (m, 5).

Removal of the side chain t-butyl group:

Z-Val-Ala-Asp(OtBu) $\text{CH}_2\text{O}$ -(6-methyl-pyrone) was dissolved under argon in 2 ml of methylene chloride add 2 mL of 50% trifluoroacetic acid methylene chloride was added and the  
15 resulting clear solution was stirred for 30 minutes. At this time silica gel TLC ( $\text{CHCl}_3$ /isopropanol:9/1) showed loss of starting material (starting material  $R_f$  0.66; product  $R_f$  0.44). The reaction was diluted twofold with chloroform and the solvents and reagents removed under reduced pressure to  
20 give a white solid, mp. 158-163°C (with multiple phase changes prior to melting).

Synthesis of

N-Morpholinecarbonyl-L-Valyl-L-Alanyl-Aspartyl

(OtBu)- $\alpha$ -(ascorbityl) methylketone. N-Morpholinecarbonyl-L-  
25 aline methyl ester: HCL-Valine methyl ester (25 g) was dissolved under argon in 600 mL of freshly distilled THF and 100 mL of dry DMF and 1.0 equivalents of N-methyl morpholine. The resulting solution was cooled to -15° and an additional 1.1 equivalent of N-methyl morpholine followed by  
30 1.1 equivalents of morpholine chloride was added. The reaction was allowed to come slowly to room temperature and stir overnight. The reaction is then poured into 300 mL of 1N HCL and extracted with ethyl acetate (2 x 200 mL). The combined organic fractions were washed with 1N HCL (50 mL),



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brine (50 mL), dried over  $\text{MgSO}_4$ , filtered and the solvents were removed under reduced pressure to give 33 g of N-morpholinecarbonyl-valine methyl ester as a white solid: NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  0.9 (dd, 6), 2.1 (septet, 1), 3.0 and 3.65 (morpholine triplets, 4 and 4), 4.98 (N-H).

Conversion to Mu Val OH: The above methyl ester was dissolved in 300 mL of methanol and 157 mL of 1N aqueous sodium hydroxide was added. The reaction was stirred at room temperature for 2 hrs. after which time TLC showed the product as a stationary spot. The methanol was removed under reduced pressure and an additional 28 mL of 1N aqueous sodium hydroxide was added and the water fraction was washed with ethyl acetate (75 mL). The water fraction was then acidified with 185 mL of 1N HCl, 3/4 of the water was removed under reduced pressure and the resulting mixture was extracted with ethyl acetate (2 x 300 mL). The organic fraction was washed with 1N HCl (50 mL), brine (50 mL), dried over  $\text{MgSO}_4$ , filtered and concentrated to give N-morpholinecarbonyl-valine 29.75 g (78% yield).

Condensation with Alanine methyl ester: Mu-Val-OH was dissolved in 300 mL of freshly distilled THF under argon and the solution was cooled to  $-15^\circ\text{C}$ . Next one equivalent of N-methyl morpholine followed by one equivalent of isobutylchloroformate was added. The reaction was allowed to activate 20 minutes and then another equivalent of N-methyl morpholine followed by alanine methyl ester hydrochloride salt was added. The reaction was allowed to come slowly to room temperature and to stir overnight. The next day the reaction was poured into 1N hydrochloric acid and extracted with ethyl acetate (2 x 200 mL). The combined organic layers were washed with 1N hydrochloric acid (50 mL), brine, dried over  $\text{MgSO}_4$ , filtered, and the solvents were removed under reduced pressure to give 10.74 g (78%) of N-morpholinecarbonyl-valyl-alanyl methyl ester. NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  1.4 (d, Ala  $\text{CH}_3$ ), 3.7 (s, OMe) ppm.

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Hydrolysis to the Free Acid: Mu-Val-Ala-OMe (1 g) was dissolved in 15 mL of methanol and then 4.8 mL of 1N sodium hydroxide (aq) was added. The reaction was allowed to continue until TLC ( $\text{CHCl}_3$ /isopropanol:9/1) showed only a stationary spot. The methanol was then removed under reduced pressure and an additional 1.2 mL of ethyl acetate and the water fraction was acidified with 6 mL of 1N HCl. The mixture is extracted with about 50 mL of ethyl acetate and the organic fraction washed with 5 mL of 1N HCl, dried over  $\text{MgSO}_4$ , filtered and concentrated to give 0.8 g (84%) of a white solid which was characterized by the loss of the NMR absorption at 3.7 ppm.

Condensation with Asp(OtBu)OH: Asp(OtBu)OH (2.51 g) was dissolved in 40 mL of dry DMF under argon and 8.2 mL of bis(trimethylsilyl)acetamide (BSA) and the reaction was allowed to stir 40 minutes. In a separate flask, Mu-Val-Ala-OH (4.0 g) was dissolved in 200 mL of dry THF under argon and the resulting solution was cooled to  $-15^\circ\text{C}$  and one equivalent of N-methylmorpholine was added followed by one equivalent of isobutylchloroformate and the resulting mixture was allowed to stir 20 minutes and then the first reaction was then poured into the second reaction and both were maintained at  $-15^\circ\text{C}$  for one hour and then allowed to come slowly to room temperature and to stir overnight. The reaction was poured into 150 mL of 1N HCl and extracted with 2 x 200 mL of ethyl acetate. The combined organic fractions were washed with 15 mL 1N HCl, brine (50 mL), dried over  $\text{MgSO}_4$  (with decolorizing carbon), filtered, and the solvents were removed under reduced pressure to yield 4.89 g of Mu-Val-Ala-Asp(OtBu)OH.

Conversion to the diazoketone: Mu-Val-Ala-Asp(OtBu)OH (4.89 g) was dissolved in 250 mL of freshly distilled THF under argon and the resulting solution was cooled to  $-15^\circ\text{C}$ . Next one equivalent of N-methyl morpholine followed by one equivalent of isobutyl chloroformate was added. The reaction

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was allowed to activate at this temperature for 20 minutes and then poured through a filter into a solution of diazomethane in ether that was made from 10.8 of diazald according to the supplier's (Aldrich) directions. The  
5 reaction was allowed to come slowly to room temperature and to stir overnight. The next day the reaction was washed with water, bicarbonate and brine (50 mL each), dried over  $\text{MgSO}_4$ , filtered and the solvents were removed under reduced pressure to give a yellow oil showing five spots on TLC  
10 (silica gel,  $\text{CHCl}_3$ /isopropanol:97/3). The lowest  $R_f$  is isolated by chromatography on 300 g of silica gel and is shown to be the product by the  $\text{CHN}_2$  absorption in the NMR at  $\delta$  5.75.

Conversion to the bromoketone: Mu-Val-Ala-Asp(OtBu) $\text{CHN}_2$   
15 was dissolved in 45 mL of methylene chloride and the resulting solution was cooled to  $-15^\circ\text{C}$ . Next 1.7 ml of 30% methylene chloride dissolved in 30 ml methylene chloride was added dropwise and the reaction was monitored by TLC (silica gel,  $\text{CHCl}_3$ /isopropanol). The reaction was then poured into  
20 brine and the organic fraction was washed with sodium bicarbonate (aq), brine, and dried over  $\text{MgSO}_4$ , filtered, and the solvents were removed under reduced pressure to give a crude gold solid which was purified by dissolving the material in a minimum of methylene chloride and precipitation  
25 in ether/hexane. The product Mu-Val-Ala-Asp(OtBu) $\text{CH}_2\text{Br}$  is characterized in the NMR (100 MHz,  $\text{CDCl}_3$ ) by the disappearance of the diazoketone absorbance at  $\delta$  5.75 and the appearance of a singlet at  $\delta$  4.18.

ICE Inhibitors: Mu-Val-Ala-Asp(OtBu) $\text{CH}_2\text{Br}$  (0.36 mmol),  
30 potassium fluoride (1.09 mmol) and the hydroxy heterocycle (0.546 mmol) was sealed under argon and then 8 mL of dry DMF was added and the reaction was allowed to stir overnight. The next day the reagents were removed either by dilution with ethyl acetate and washing brine or by passage through  
35 silica gel. The solvents were removed under vacuum and the

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product was isolated by size exclusion chromatography (LH20). In this manner the following compounds were prepared:

N-Morpholinecarbonyl-L-Valyl-L-Alanyl-Aspartyl(OtBu)- $\alpha$ -(ascorbityl) methylketone (mp. 138-144°C);

5 N-Morpholinecarbonyl-L-Valyl-L-Alanyl-Aspartyl(OtBu)- $\alpha$ -(-4-oxy-(3-azo-m-anisidine) methyl ketone: NMR (CDCl<sub>3</sub>)  $\delta$  0.95 (dd, 6H, Val CH<sub>3</sub>), 1.4 (s, 12H, OtBu+Ala CH<sub>3</sub>), 2.1 (m, 1H, Val CH), 2.9 (d, 2H, CH<sub>2</sub> sidechain), 3.3 (t, 4H, MU), 3.7 (t, 4H, MU), 3.85  
10 (s, 3H, OCH<sub>3</sub>), 4.2 (t, 1H), 4.6 (t, 1H), 4.7 (d, 2H), 4.9 (m, 3H, CH<sub>2</sub>O), 6.9 (d, 1H), 7.1 (m, 4H) 7.9 (d, 1H).

Removal of the tBu in the above inhibitors was effected with 25% trichloroacetic acid in methylene chloride to give the corresponding free acid inhibitors.

15

## EXAMPLE 11

## SYNTHESIS OF CALPAIN INHIBITORS

The following example is meant to be illustrative of a calpain inhibitor but is not meant to be restrictive as numerous variations in peptide and leaving groups of this  
20 invention can be envisioned without undue experimentation.

Acetyl-Leucyl-Leucyl-Phenylalanyl- $\alpha$ -(-4-oxy-dihydrofuran-2-one) methyl ketone. Ac-Leu-Leu-OCH<sub>3</sub>: Acetyl-Leucine (5.0 g) was dissolved in 150 mL of distilled THF under argon and the resulting solution was cooled to -15°. Next one  
25 equivalent of N-methyl morpholine followed by one equivalent of isobutyl chloroformate was added and the reaction was allowed to activate 20 minutes and then another equivalent of N-methyl morpholine followed by HCl-LeuOMe (5.25 g). The reaction is allowed to slowly come to room temperature and  
30 stir overnight. The reaction was then poured into 150 mL of 1N HCl and extracted with ethyl acetate (2 x 150 mL). The combined organic fractions were washed with 1N HCl (15 mL),

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brine (50 mL), and dried over  $\text{MgSO}_4$ , filtered and the solvent was removed under reduced pressure and then high vacuum. TLC (silica gel,  $\text{CHCl}_3$ /isopropanol:95:5) showed the product Ac-Leu-Leu- $\text{OCH}_3$  to be a single spot  $R_f$  0.36.

5 NMR (100 MHz)  $\delta$  0.95 (d, 12H), 1.5 (bs, 6H), 2.0 (s, 3H), 3.7 (s, 3H), 4.5 (q, 2H), 6.6 (d, 1H), 6.8 (d, 1H).

Hydrolysis to the Free Acid: Ac-Leu-Leu- $\text{OCH}_3$  (7.8 g) was dissolved in 150 mL of methanol and then 38 mL of 1N sodium hydroxide was added and the reaction was allowed to stir at room temperature for about 4 hours. The methanol was removed under reduced pressure and an additional 10 mL of 1N sodium hydroxide was added and the water fraction was extracted with 10 mL of ethyl acetate. The water fraction was then neutralized with 1N HCl and extracted with ethyl acetate (3 x 50 mL). The combined organic fraction was washed with brine and dried over  $\text{MgSO}_4$ , filtered and the solvent was removed under reduced pressure.  $\delta$  0.95 (d, 12H), 1.5 (br s, 6H), 2.0 (s, 3H), 4.5 (q, 2H), 6.6 (d, 1H), 6.8 (D, 1H), 9.5 (bs, 1H).

20 Condensation with PheOMe. Ac-Leu-Leu-OH was dissolved in 200 mL of distilled THF under argon and the solution was cooled to  $-15^\circ\text{C}$ . Next one equivalent of N-methyl morpholine followed by one equivalent of isobutyl chloroformate was added and the reaction was allowed to activate for 20 minutes. An additional equivalent of N-methyl morpholine followed by HCl-HPhe $\text{OCH}_3$  was added and the reaction was allowed to come slowly to room temperature and stir overnight. The reaction was poured into 200 mL of 1N HCl and extracted with ethyl acetate (2 x 150 mL). The organic fraction was washed with 1N HCl (20 mL), brine (50 mL), and dried over  $\text{MgSO}_4$ , filtered and the solvents removed under reduced pressure and then high vacuum to leave a solid white cake (TLC, silica gel,  $\text{CHCl}_3$ /isopropanol  $R_f$  0.35). NMR (100 MHz)  $\delta$  0.95 (d, 12H), 1.6 (bs, 6H), 2.1 (s, 3H), 3.1 (d, 2H), 3.6 (s, 3H), 4.7 (m, 3H), 6.9 (d, 1H), 7.2 (m, 5H), 7.5 (d, 1H).

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Conversion to the Free Acid. Ac-Leu-Leu-Phe-OMe was dissolved in 150 mL of methanol and 26 mL of 1N sodium hydroxide was added and the reaction was stirred 4 hours at which time the methanol was removed under reduced pressure and an additional 7 mL of sodium hydroxide was added. This water fraction was then washed with ethyl acetate (10 mL) and neutralized by the addition of 1N HCL. The resulting mixture was extracted with ethyl acetate 2 x 100 mL and the extract washed with 1N HCL, brine and dried over  $\text{MgSO}_4$ , filtered and the solvents were removed under reduced pressure to give 7.56 g (94%) of Ac-Leu-Leu-Phe-OH as a white solid. The NMR (100 MHz,  $\text{CDCl}_3$ ) of the product acid bears striking resemblance to that of the precursor ester except for the loss of a signal at  $\delta$  3.6 and the appearance of a broad singlet at 10.1 (1H).

Conversion to the Diazoketone: Ac-Leu-Leu-Phe-OH (4.68 g) was dissolved in 200 mL of freshly distilled THF and a methanol-ice bath was applied. Next one equivalent of N-methyl morpholine followed by one equivalent of isobutyl chloroformate was added and the reaction was allowed to activate for 20 minutes and then the resulting mixture was poured through filter paper into diazomethane/ether made from 10.8 g of Diazald according to the supplier's (Aldrich) directions. The reaction was allowed to come slowly to room temperature and to stand overnight. The reaction was washed with water (2 x 50 mL), sodium bicarbonate (50 mL), brine (50 mL), and then dried over  $\text{MgSO}_4$ , filtered and the solvents were removed under reduced pressure to give after column chromatography (silica gel,  $\text{CHCl}_3$ /isopropanol:93/7) 3.07 g (61%) of a yellow powder. NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  0.95 (d, 12H), 1.6 (bs, 6H), 2.1 (s, 3H), 3.1 (d, 2H), 4.7 (m, 3H), 5.6 (s, 1H), 6.9 (d, 1H), 7.2 (m, 5H), 7.5 (d, 1H).

Conversion to the Bromide. Ac-Leu-Leu-Phe- $\text{CHN}_2$  (1 g) was dissolved in 175 mL of methylene chloride and then 1.2 mL of 30% HBr/acetic acid that had been diluted with 25 mL

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methylene chloride was added dropwise at -15°C. As the reaction proceeds bubbles evolve with the formation of a precipitate. The reaction was monitored by TLC (silica gel/CHCl<sub>3</sub>-isopropanol: 9/1; R<sub>f</sub> product 0.54). Upon  
5 completion, the reaction was poured into 150 mL of brine and the reaction flask was washed with another 150 mL of methylene chloride to dissolve the residual precipitate. The combined organic layers were washed with sodium bicarbonate (aq, 50 mL), brine (50 mL), dried over MgSO<sub>4</sub> and  
10 concentrated to give a dull white solid. This solid was purified by precipitation from methylene chloride into ether to yield 610 mg (54%) of a white solid TLC (silica gel, CHCl<sub>3</sub>/isopropanol:9:1) one spot R<sub>f</sub> 0.54. NMR (DMSO-d<sub>6</sub>)  
15 δ 0.81 (d, 12H), 1.3 (m, 6H), 1.8 (s, 3H), 3.1 (d, 2H), 4.1 (m, 2H), 4.3 (s, 2H), 4.6 (q, 1H), 7.2 (m, 5H), 8.0 (d, 2H), 8.4 (d, 1H).

Calpain inhibitor: Ac-Leu-Leu-Phe-CH<sub>2</sub>Br (200 mg), tetronic acid (65 mg) and potassium fluoride (68 mg) were mixed under argon with 5 mL of dry DMF overnight. The  
20 reaction was then diluted with 20 mL of ethyl acetate and the reaction was washed with 10 mL sodium bicarbonate (aq), brine (10 mL), and dried over MgSO<sub>4</sub>. The reaction was filtered and the solvents were removed under reduced pressure and then high vacuum to give 107 mg (49%) of  
25 acetyl-leucyl-leucyl-phenylalanyl-α-(4-oxy-dihydrofuran-2-one)methyl ketone.

EXAMPLE 12  
SYNTHESIS OF OTHER HETEROCYCLIC INHIBITORS

Using the following procedures other heterocyclic  
30 cathepsin inhibitors are prepared.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine-α-(4-oxy-N-acetyl-proline methyl ester) methyl ketone (A1).

MuPheHPheCH<sub>2</sub>Br (250 mg), N-acetyl-proline methyl ester

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(2.0g), potassium fluoride (232 mg), and potassium carbonate (276 mg) are placed under argon and then 1.5 mL of dry DMF was added and the reaction was allowed to stir at room temperature for 100 minutes. The reaction was then passed  
5 through a short silica gel column (ethyl acetate) and the solvents were removed in vacuo. Precipitation in ether produced a white solid, mp. 81-84°C.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanyl- $\alpha$ -(3-oxy-5-ethyl-4-methyl 2(5H) furanone) methyl ketone (A2).  
10 MuPheHPheCH<sub>2</sub>Br (100 mg), potassium fluoride (45 mg), and 5-ethyl-3-hydroxy-4-methyl-2(5H) furanone (110 mg) was placed under argon in 5 mL of dry DMF and the reaction was stirred at room temperature overnight. The next day the reaction was diluted with ethyl acetate and washed with aqueous sodium  
15 bicarbonate and the brine, dried over MgSO<sub>4</sub>, filtered and the solvents were removed in vacuo. The product was purified by size exclusion chromatography (LH-20, methanol) to give a white solid, mp. 65-71°C.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(8-oxy-quinoline) methyl ketone (A3). To MuPheHPheCH<sub>2</sub>Br  
20 (100 mg), potassium fluoride (45 mg) and 8-hydroxyquinoline (123 mg) in a test tube under argon was added 5 mL of dry DMF and the reaction was allowed to stir for four hours. The reaction was then passed through a short silica gel column  
25 and the solvents were removed in vacuo. The product was purified by first size exclusion chromatography (LH-20, methanol) and then by crystallization from methylene chloride/ether to give 65 mg of crystals. The product was characterized by NMR (100 MHz)  $\delta$  8.5-8.0 (m, hetero  
30 aromatic), 7.5-6.5 (mm, homo and heteroaromatic), 3.75-3.5, 3.25-3.0 (2m, Mu H), 2.75 (s, heteroaromatic Me) ppm.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(2-oxy-4-methyl-quinoline) methyl ketone (A4).  
MuPheHPheCH<sub>2</sub>Br (100 mg), potassium fluoride (45 mg), and  
35 2-hydroxy-4-methyl-quinoline (123 mg) was placed in a test



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tube under argon and 5 mL of dry DMF was added. The reaction was allowed to stir at room temperature overnight. The reaction was then passed through a short silica gel plug and the solvents were removed in vacuo. The residue was purified first by size exclusion chromatography and then by precipitation into ether to give a solid, mp. 180-183°C.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(4-oxy-quinoline) methyl ketone (A5). MuPheHPheCH<sub>2</sub>Br (100 mg), potassium fluoride (45 mg), and 4-hydroxyquinoline was placed in a test tube under argon and 5 mL of dry DMF was added. The reaction was stirred for 3.5 hours and then passed through a short silica gel column (ethyl acetate). The solvents were removed in vacuo and the residue was purified by size exclusion chromatography followed by precipitation of the collected product in ether to yield a white powder, mp. 107-111°C.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(4-oxy-quinazoline) methyl ketone (A6). MuPheHPheCH<sub>2</sub>Br (100 mg), 5-methyl-5-triazolo[1,5a]-pyrimidin-7-ol (116 mg), and potassium fluoride (45 mg) was added together under argon in a dry test tube and 5 mL of dry DMF was added. The reaction was stirred at room temperature for 3.5 hours and then the reaction was diluted with ethyl acetate and passed through a plug of silica gel. The solvents were removed in vacuo and the product was purified by size exclusion chromatography (LH-20, methanol) to give a solid product, mp. 129-132°C.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(2-oxy-benzimidazole) methyl ketone (A7). MuPheHPheCH<sub>2</sub>Br (100 mg), 2-hydroxybenzimidazole (104 mg), and potassium fluoride (45 mg) was added together under argon in a dry test tube and 5 mL of dry DMF was added. The reaction was stirred at room temperature for 3 hours and then the reaction was diluted with ethyl acetate and passed through a plug of

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silica gel. The solvents were removed in vacuo and the product was purified by size exclusion chromatography (LH-20, methanol) to give an off white solid product, mp. 115-120°C.

5 N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(1-oxy-isoquinoline) methyl ketone (A8). MuPheHPheCH<sub>2</sub>Br (100 mg), potassium fluoride (45 mg), isocarbostyrl (112 mg), are placed under argon and then 4 mL of dry DMF is added. The reaction is stirred at room temperature for three hours and then the reaction is diluted with ethyl acetate and  
10 passed through a short silica gel column. The solvents are removed in vacuo and the product purified by size exclusion chromatography (LH-20, methanol) to give a white solid, mp. 104-107°C.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(7-oxy-coumarin) methyl ketone (A10). MuPheHPheCH<sub>2</sub>Br (200 mg) and potassium fluoride (90 mg) was added under argon to 1.5 mL of DMF and 250 mg of 7-hydroxycoumarin was added. The reaction turns a bright gold and is allowed to stir for one hour at which time TLC shows total loss of bromide. The  
20 reaction was then passed through a short column of silica gel (CHCl<sub>3</sub>/isopropanol, 9:1) and the solvents were removed in vacuo. Further chromatography (LH-20/methanol) gave after removal of solvent a white solid foam, mp. 87-89°C.

In a like manner N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanyl- $\alpha$ -(7-oxy-4-methyl-coumarin) methyl ketone (A9) was prepared, mp. 99-102°C.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(2-oxy-benzofuran) methyl ketone (A11). MuPheHPheCH<sub>2</sub>Br (100 mg), 2-coumaranone (52 mg) and potassium fluoride (45 mg) were placed in a test tube under argon and then one mL of DMF was added and the reaction turns a cherry red. The reaction after 20 minutes shows a loss of starting bromide (TLC, Silica gel, CHCl<sub>3</sub>/isopropanol:9/1) R<sub>f</sub> product, 0.59; R<sub>f</sub> bromide 0.48. The reaction was passed through a  
30

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short plug of silica gel (ethyl acetate) and the solvents were removed in vacuo. The residue was dissolved in a minimum of methylene chloride and precipitated in ether and the precipitate filtered to yield a white solid, mp. 94-110°C.

5     N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(3-oxy-2-methyl-4-pyrone) methyl ketone (A12).

MuPheHPheCH<sub>2</sub>Br (94 mg), potassium fluoride (45 mg), and 3-hydroxy-2-methyl-4-pyrone was placed in a test tube under argon and 5 mL of dry DMF was added and the reaction was  
10     stirred at room temperature for two hours at which time the reaction showed a loss of starting material (silica gel, CHCl<sub>3</sub>/isopropanol, 9/1). The reaction was then passed through a short plug of silica gel and the solvents were removed in vacuo. The product was then purified by size  
15     exclusion chromatography to give after evaporation of solvent a gold solid, mp. 71-81.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(2-oxy-benzothiazole) methyl ketone (A13). MuPheHPheCH<sub>2</sub>Br (100 mg), 2-hydroxybenzothiazole (117 mg), and potassium  
20     fluoride (45 mg) were placed in a test tube under argon and 3 mL of dry DMF was added. The reaction was stirred at room temperature until TLC (silica gel) showed loss of starting material. The reaction was passed through a short silica gel column, the solvents were removed in vacuo. The residue is  
25     dissolved in hot methanol and a white precipitate forms which upon filtration proves to be the product, mp. 211-213°C.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(2-oxy-thiophene) methyl ketone (A14). MuPheHPheCH<sub>2</sub>Br (265 mg), potassium carbonate (119 mg), and potassium  
30     fluoride (284 mg) were added together under argon and then one gram of thiophenone in 4 mL DMF was added and the reaction was allowed to stir at room temperature for 2 hours. The solvents were removed in vacuo and the products

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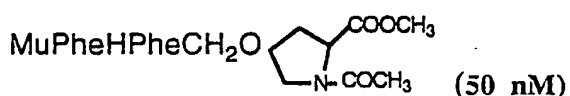
were separated on a 10g silica gel column.

N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -  
-(5-oxy-3-methyl-4-isoxazolecarboxylate) methyl ketone  
(A15). MuPheHPheCH<sub>2</sub>Br (510 mg), ethyl

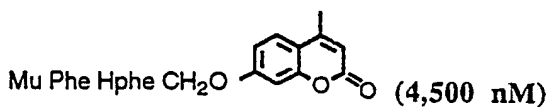
5 5-hydroxy-3-methyl-4-isoxazole carboxylate sodium salt, and 5  
mL of DMF was allowed to stir under argon at room temperature  
for 4 hours. The reaction was then passed through a short  
plug of silica gel (CHCl<sub>3</sub>/isopropanol) and the solvents  
were removed in vacuo. The product was purified by size  
10 exclusion chromatography (LH-20) to give after precipitation  
in ether and filtration a white solid that melted with a  
phase change at 98-105 and then again at 125-131°C.

Formation of alkyl halide salts from inhibitors  
containing nitrogen in the heterocycle leaving group: The  
15 methyl iodide isoquinoline salt of  
N-Morpholinecarbonyl-L-phenylalanyl-L-homophenylalanine- $\alpha$ -(1  
-oxy-isoquinoline) methyl ketone (A16). Compound A8 (83 mg)  
is dissolved in 2 mL of toluene and one mL of iodomethane is  
added. The reaction is sealed and allowed to stir for two  
20 days. A white precipitate forms which is filtered and dried  
under vacuum to give a white solid, mp. 159-161°C.

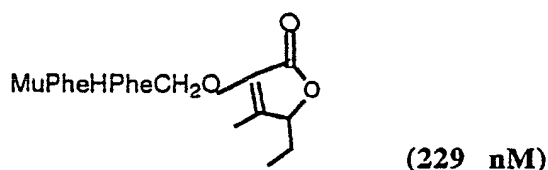
**CONSTRUCTIONS OF INHIBITORS WITH OTHER HETEROCYCLES**  
**(IC<sub>50</sub> Cathepsin B Inhibition)**



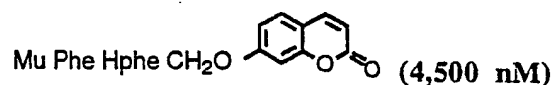
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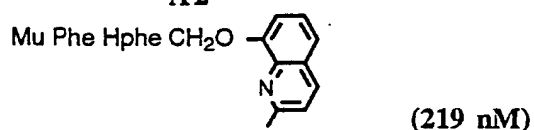
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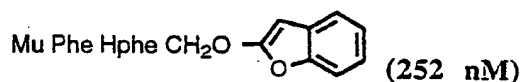
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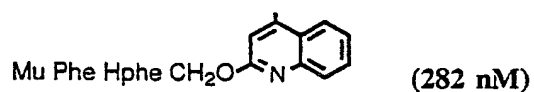
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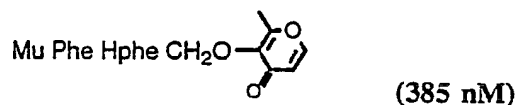
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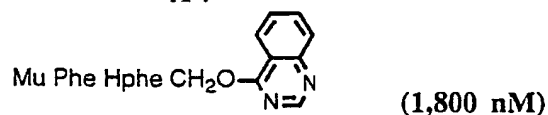
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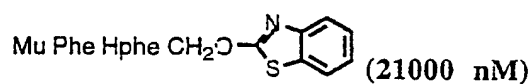
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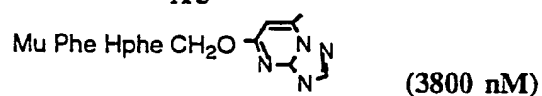
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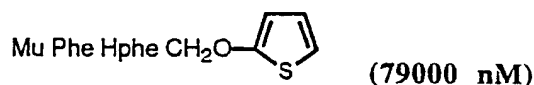
A 5



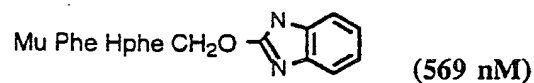
A 13



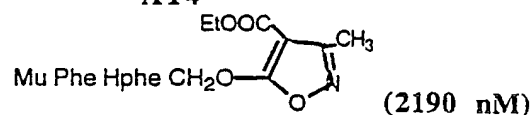
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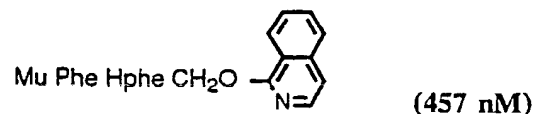
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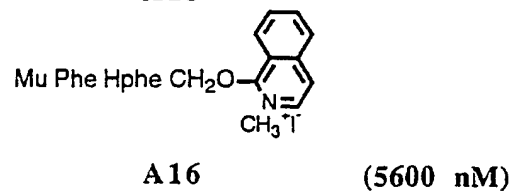
A 7



A 15



A 8



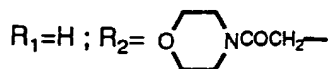
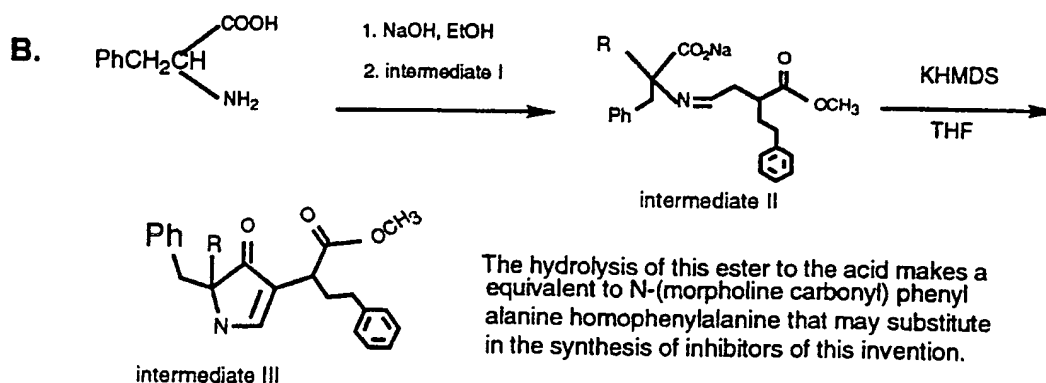
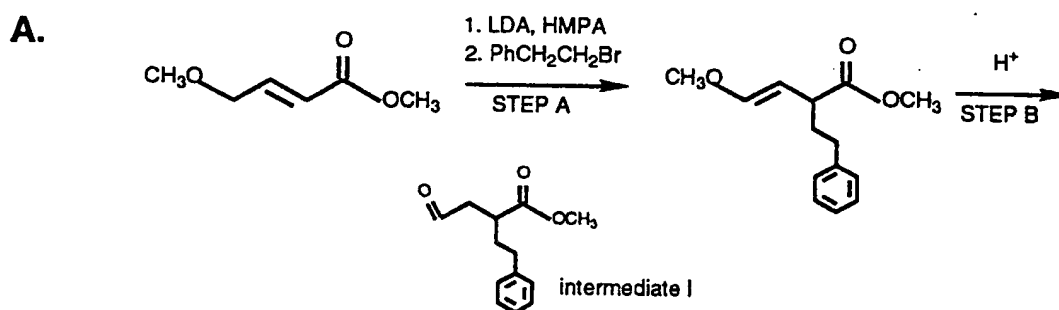
A 16

## EXAMPLE 13

Synthesis of Inhibitors With Heterocycles  
in their Peptide Backbones

The reaction scheme shown below is a first method for  
 5 the synthesis of cysteine protease inhibitors with  
 heterocycles in their peptide backbones. The synthetic  
 method is an adaptation from that of Amos B. Smith and Ralph  
 Hirshman as disclosed in "Design and Synthesis of  
 Peptidomimetic Inhibitors of HIV-1 Protease and Renin," 37  
 10 J. Med. Chem. 215.

## Synthesis of heterocycles in peptide backbone : Method I\*

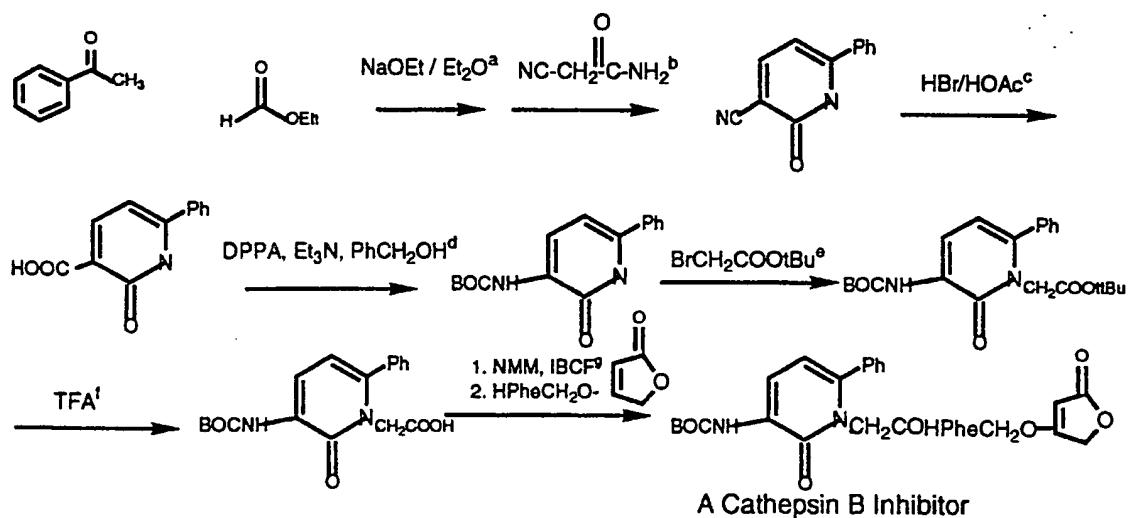


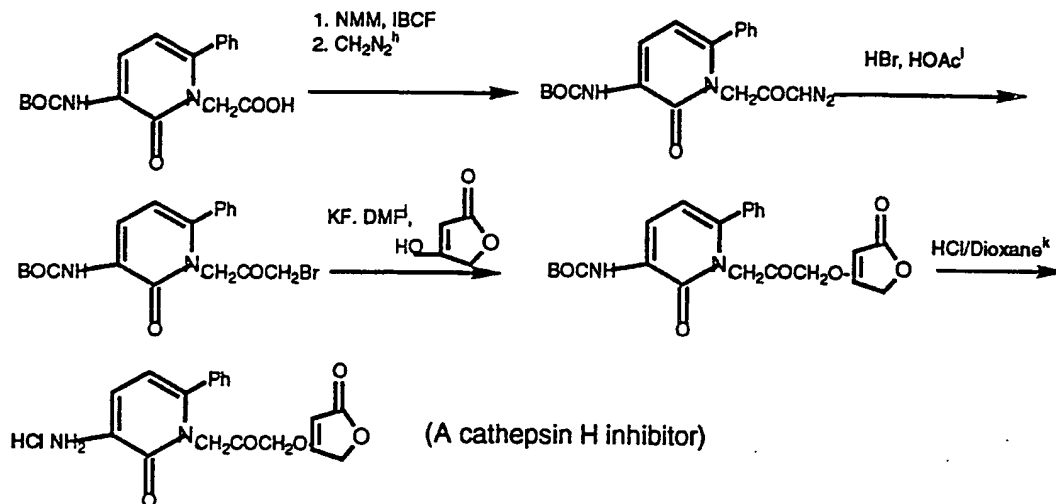
-61-

The reaction schemes shown below illustrate a second method for the synthesis of cysteine protease inhibitors with heterocycles in their peptide backbones. This synthetic method is an adaptation from that of Damewood et al., "Nonpeptidic Inhibitors of Human Leukocyte Elastase," 37 *J. Med. Chem.* 3303.

### Synthesis of heterocycles in peptide backbone : Method 2\*\*

#### A. Heterocycle at P<sub>2</sub>



**B. Heterocycle at P<sub>1</sub>****EXAMPLE 14****Protocol for Testing ICE Inhibitors**

The percent inhibition of two inhibitors,  
 N-Benzoxycarbonyl-Valyl-Alanyl-Aspartyl- $\alpha$ -(4-oxy-(6-methyl-  
 2-pyrone) methyl ketone, and N-Morpholinecarbonyl-L-  
 Valyl-L-Alanyl-Aspartyl (OtBu)- $\alpha$ -(ascorbityl) methyl  
 ketone, on IL-1 $\beta$  protease was determined as follows. A 10mM  
 dithiothreitol, 100 mM Hepes, 10% sucrose, 0.1% CHAPS, pH  
 7.5 buffer solution with 50 $\mu$ M Z-YVAD-AFC substrate was  
 prepared. The enzyme was activated for 1 minute in the  
 buffer/substrate solution at room temperature. Inhibitor



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was prepared as stock solution in dimethyl sulfoxide. Inhibitor and enzyme/buffer were incubated for 15 minutes at 37C. Final concentrations of inhibitor were 2000nM, 200nM, and 20nM. Enzyme activity was followed by the release of  
5 free fluorescent detecting group over sixty minutes at 37C, as compared to the control.

## EXAMPLE 15

## Protocol for Testing Calpain Inhibitors

The percent inhibition of one inhibitor, Acetyl-Leucyl-  
10 Leucyl-Phenylalanyl- $\alpha$ -(4-oxy-dihydrofuran-2-one) methyl ketone, on calpain (Calcium Activated Neutral Protease) was determined as follows. A 50mM Hepes, 10mM calcium chloride, 5mM cysteine, 1mM  $\beta$ -mercaptoethanol, pH 7.5 buffer solution was prepared. The enzyme was activated for 1 minute in the  
15 buffer solution at room temperature. Inhibitor was prepared as stock solution in dimethylformamide. Inhibitor and enzyme/buffer were incubated for 30 minutes at 37°C. Final concentrations of inhibitor were 20 $\mu$ M, 2 $\mu$ M, and 200nM. Enzyme activity was followed with 200 $\mu$ M  
20 Boc-Valnyl-Leucyl-Lysine-AFC substrate by the release of free fluorescent detecting group over minutes at 37°C, as compared to the control. The inhibitor showed activity against the enzyme at less than 2 $\mu$ M.

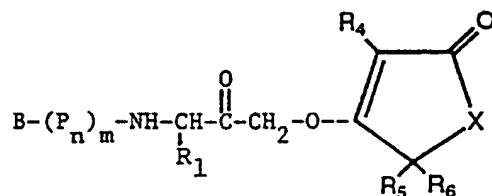
While the invention has been illustrated and described  
25 in detail in the drawing and foregoing description, the same is to be considered as illustrative and not restrictive in character, it being understood that only the preferred embodiment has been shown and described and that all changes and modifications that come within the spirit of the  
30 invention are desired to be protected.

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## CLAIMS

What is claimed is:

1. Cathepsin or calpain inhibitors of the formula:



where B is H or an N-terminal blocking group;

5  $R_1$  is the amino acid side chain of the  $P_1$  amino acid residue; wherein the  $P_1$  amino acid is not Asp; each  $P_n$  is an amino acid residue, or is a heterocyclic replacement of the amino acid wherein the heterocycle is a piperazine, a decahydroisoquinoline, a pyrrolinone, a pyridine, a pyridone, a carbolinone, a quinazoline, a pyrimidone or the like;

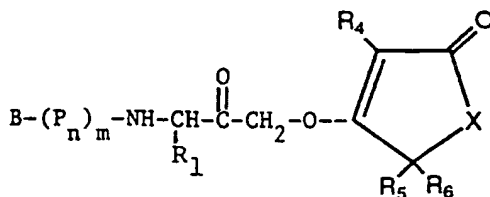
m is 0 or a positive integer;

$R_4$  is a hydroxyl, alkoxyl, acyl, hydrogen, alkyl or phenyl;

15  $R_5$  and  $R_6$  are jointly a carboxyl group or a double bond terminating in an alkyl or an aryl group, or are independently acyl, aryl or heteroaryl if  $R_4$  is hydrogen, alkyl or phenyl, or are independently acyl, alkyl, hydrogen, aryl or heteroaryl otherwise; and

20 X is N, S, O or  $CH_2$ .

2. ICE inhibitors of the formula:

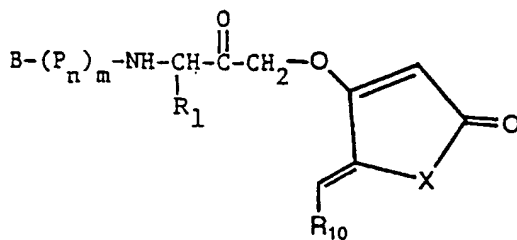


-65-

where B is H or an N-terminal blocking group;  
 $R_1$  is the Asp amino acid side chain;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 5 the heterocycle is a piperazine, a  
 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 10  $R_4$  is a hydroxyl, alkoxy, acyl, hydrogen, alkyl  
 or phenyl;  
 $R_5$  and  $R_6$  are jointly a carboxyl group or a  
 double bond terminating in an alkyl or an aryl  
 group, or are independently acyl, aryl or heteroaryl  
 15 if  $R_4$  is hydrogen, alkyl or phenyl, or are  
 independently acyl, alkyl, hydrogen, aryl or  
 heteroaryl otherwise; and  
 X is N, S, O or  $CH_2$ .

3. An ICE inhibitor of claim 2 wherein X is  $CH_2$  and  
 20  $R_5/R_6$  is a carbonyl.

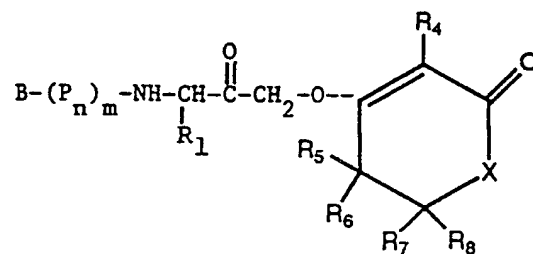
4. Cysteine protease inhibitors of the formula:



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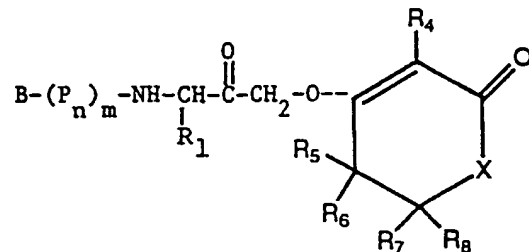
where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 5 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 $R_{10}$  is H or an optionally substituted alkyl, aryl,  
 10 heteroaryl, or the residue of a sugar; and  
 X is N, S, O or  $CH_2$ .

5. Cysteine protease inhibitors of the formula:



where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 15 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 20 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxyl, acyl, hydrogen, alkyl  
 or phenyl;  
 $R_5$ ,  $R_6$ ,  $R_7$  and  $R_8$  are independently  
 hydrogen, alkyl, acyl, phenyl, halo, hydroxyl, oxy  
 25 or alkoxy; and  
 X is N, S, O or  $CH_2$ .

## 6. Cysteine protease inhibitors of the formula:



where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;

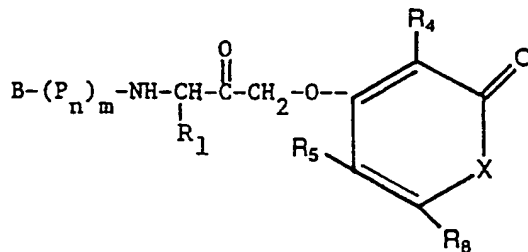
m is 0 or a positive integer;

$R_4$  is a hydroxyl, alkoxyl, acyl, hydrogen, alkyl  
 or phenyl;

$R_5$  and  $R_6$  may be attached to  $R_7$  and  $R_8$  to  
 form a ring that is either saturated or unsaturated  
 or aromatic; and

X is N, S, O or  $CH_2$ .

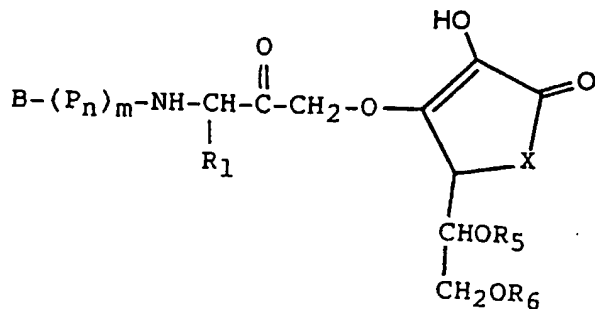
## 7. Cysteine protease inhibitors of the formula:



-68-

where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 5 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 $R_4$  is a hydroxyl, alkoxyl, acyl, hydrogen, alkyl  
 10 or phenyl;  
 $R_5$  and  $R_8$  are independently hydrogen, alkyl,  
 acyl, phenyl, halo, hydroxyl, oxy or alkoxy, or  $R_5$   
 is attached to  $R_8$  to form a homocyclic or  
 hererocyclic ring that is either saturated or  
 15 unsaturated or aromatic; and  
 X is N, S, O or  $CH_2$ .

8. Cysteine protease inhibitors of the formula:

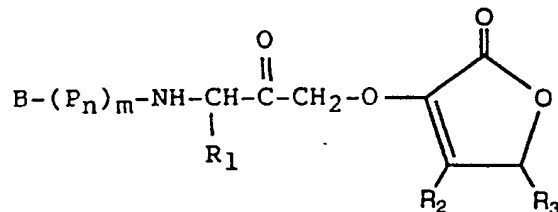


-69-

where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 5 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 $R_5$  and  $R_6$  are independently hydrogen, alkyl or  
 10 acyl; and  
 X is N, S, O or  $CH_2$ .

9. The cysteine protease inhibitors of claim 8 wherein  
 $R_5$  and  $R_6$  are each hydrogen.

10. Cysteine protease inhibitors of the formula:



15 where B is H or an N-terminal blocking group;  
 each  $P_n$  is an amino acid residue, or is a  
 heterocyclic replacement of the amino acid wherein  
 the heterocycle is a piperazine, a  
 decahydroisoquinoline, a pyrrolinone, a pyridine, a  
 20 pyridone, a carbolinone, a quinazoline, a pyrimidone  
 or the like;  
 m is 0 or a positive integer;  
 $R_2$  and  $R_3$  are independently H or an alkyl or  
 alkenyl group; and  
 25 X is N, S, O or  $CH_2$ .

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11. The cysteine protease inhibitors of claim 10 wherein  $R_2$  is  $CH_3$  and  $R_3$  is  $C_2H_5$ .

12. A cysteine protease inhibitor of claim 9 wherein the substituents of the leaving group are an extended  $C_{2-8}$  alkyl or alkenyl chain.



## INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US96/09471**A. CLASSIFICATION OF SUBJECT MATTER**

IPC(6) : Please See Extra Sheet.

US CL : Please See Extra Sheet.

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 544/283,284,291,292,293,294,322,360,372,374,379,382; 546/81,139, 275, 281, 283, 284; 548/543; 549/175, 416; 530/300

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched  
NONE

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

CAS ONLINE

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US, A, 4,518,528 (RASNICK) 21 May 1985, see entire document.	1



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents:	*T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
*A document defining the general state of the art which is not considered to be of particular relevance	*X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
*E earlier document published on or after the international filing date	*Y document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
*L document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	*G document member of the same patent family
*O document referring to an oral disclosure, use, exhibition or other means	
*P document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search

20 AUGUST 1996

Date of mailing of the international search report

04 SEP 1996

Name and mailing address of the ISA/US  
Commissioner of Patents and Trademarks  
Box PCT  
Washington, D.C. 20231

Facsimile No. (703) 305-3230

Authorized officer

CECILIA TSANG

Telephone No. (703) 308-4715

# INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US96/09471

## Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This international report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:  
because they relate to subject matter not required to be searched by this Authority, namely:
  
2. ☐ Claims Nos.:  
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
  
3. ☐ Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

## Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

Please See Extra Sheet.

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
  
4. ☒ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:  
1-12 (part), wherein Pn is amino acid and containing no additional heterocycle

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.  
☐ No protest accompanied the payment of additional search fees.

# INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US96/09471

## A. CLASSIFICATION OF SUBJECT MATTER: IPC (6):

C07D 239/72, 239/02, 401/04, 241/04, 471/04, 217/00, 409/00, 411/00, 413/00, 207/12, 207/24, 207/36, 315/00;  
C07K 2/00, 4/00, 5/00, 7/00, 14/00, 16/00, 17/00

## A. CLASSIFICATION OF SUBJECT MATTER: US CL :

544/283, 284, 291, 292, 293, 294, 322, 360, 372, 374, 379, 382; 546/81, 139, 275, 281, 283, 284; 548/543; 549/175, 416;  
530/300

## BOX II. OBSERVATIONS WHERE UNITY OF INVENTION WAS LACKING

This ISA found multiple inventions as follows:

This application contains the following inventions or groups of inventions which are not so linked as to form a single inventive concept under PCT Rule 13.1. In order for all inventions to be examined, the appropriate additional examination fees must be paid.

Group I, claim(s) 1-12 (part of each), drawn to compounds wherein Pn is amino acid containing no heterocyclic, and X=O, classified in 549/416, 175.

Group II, claim(s) 1-12 (part of each), drawn to compounds wherein Pn containing piperazine, classified in 544/360, 372, 374, 379 and 382.

Group III, claim(s) 1-12 (part of each), drawn to compounds wherein Pn containing decahydroisoquinoline, classified in 546/139+.

Group IV, claims 1-12 (part of each), drawn to compounds wherein Pn containing pyrrolinone, classified in 548/543.

Group V, claims 1-12 (part of each), drawn to compounds wherein Pn containing pyridine or pyridone, classified in 546/275, 281, 283, 284.

Group VI, claims 1-12 (part of each), drawn to compounds wherein Pn containing carbolinone, classified in 546/81.

Group VII, claims 1-12 (part of each), drawn to compounds wherein Pn containing quinazoline, classified in 544/283, 284, 291-294.

Group VIII, claim 1-12 (part of each), drawn to compounds wherein Pn containing pyrimidone, classified in class 544/322.

Group IX, claims 1-12 (part of each), drawn to compounds other than grouped above.

The inventions listed as Groups I-IX do not relate to a single inventive concept under PCT Rule 13.1 because, under PCT Rule 13.2, they lack the same or corresponding special technical features for the following reasons:

Pyrene containing no additional heterocyclic differs significantly from pyrene containing piperazine, decahydroisoquinoline, pyrrolinone or pyridine, etc. WO 93/09135 cited in the description does not consider them as functional equivalents. Prior art which may anticipated or render obvious one of the groups would not necessarily do the same for the remaining groups. Therefore, Groups I to IX do not share a single inventive concept under PCT Rule 13.1.